

Part II

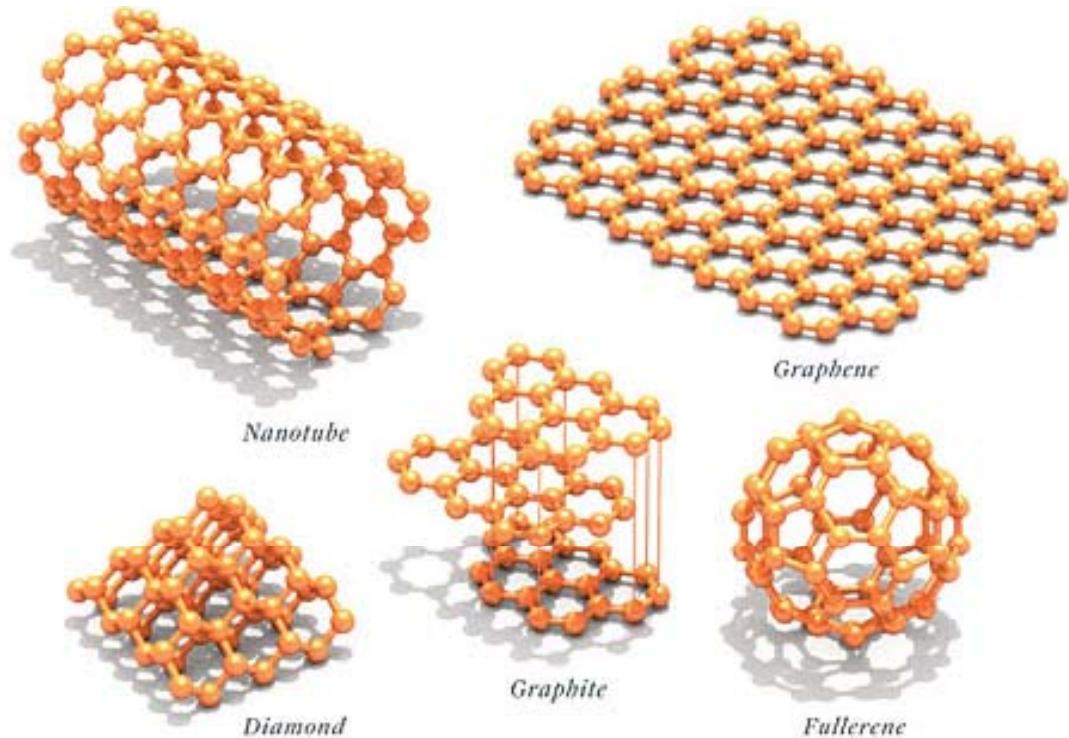
**Low-dimensional carbon materials--Carbon
nanotubes (CNTs) and Graphene**

Outline

- Carbon nanomaterials
 - Types of carbon nanomaterials
- Graphene and CNT
 - Introduction
 - Synthesis
 - Specific topics
 - Application

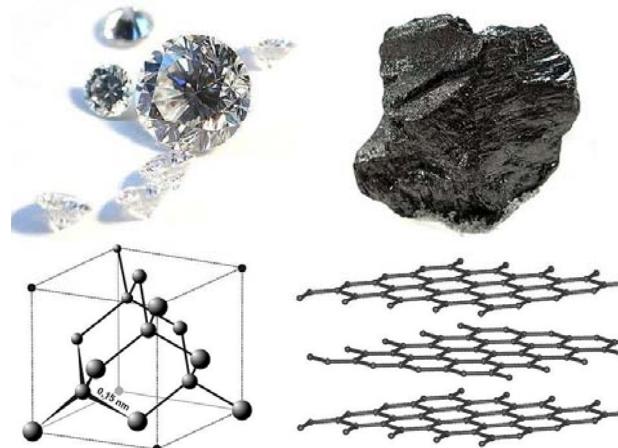
Carbon Nanomaterials

- 3D
 - Graphite
 - Diamond
- 2D
 - **Graphene**
- 1D
 - **Carbon nanotube**
- 0D
 - Fullerene

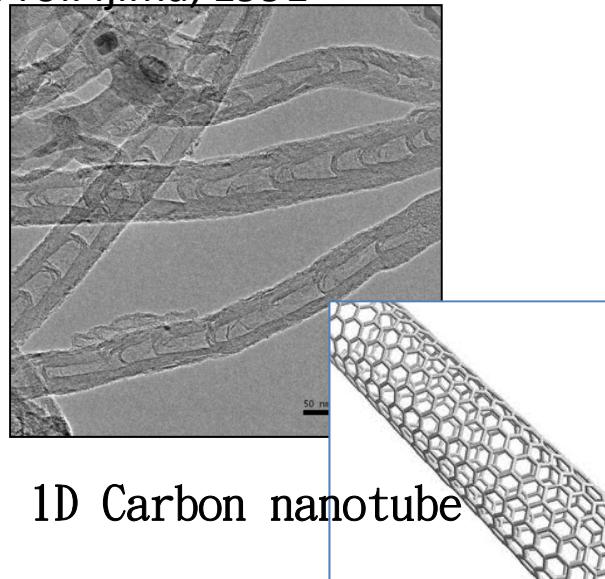


Carbon-related nanomaterials

3D Diamond and Graphite



discovered by Prof. Iijima, 1991



1D Carbon nanotube

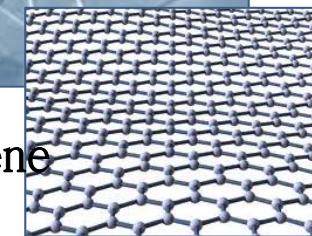
en.wikipedia.org

craigbanksresearch.com

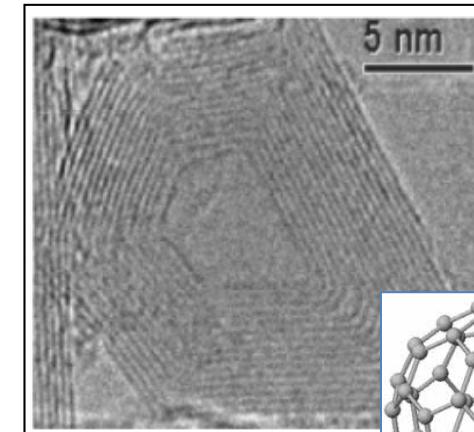
grapheneindustries.com



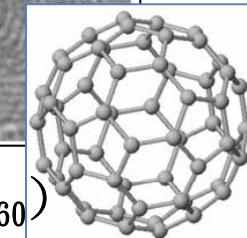
2D Graphene



(2010 Nobel
prize in physics)



0D Bulkyball(C₆₀)



(1996 Nobel priz
in Chemistry)

2-1. Introduction of CNT

CNT History

- 1952: Radushkevich and Lukyanovich found nano-sized carbon fibers (Russian)
- 1991: CNTs were discovered by Prof. Iijima and named
- 1992: Theoretical predictions of the electronic properties of SWCNT
- 1997: First CNT transistor

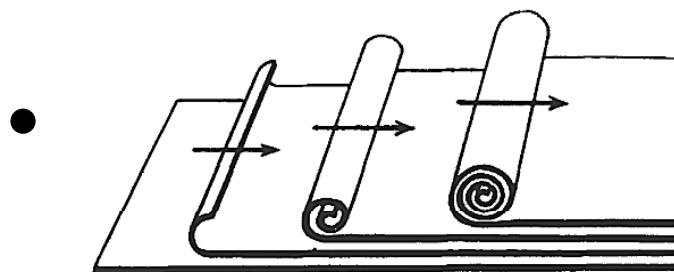
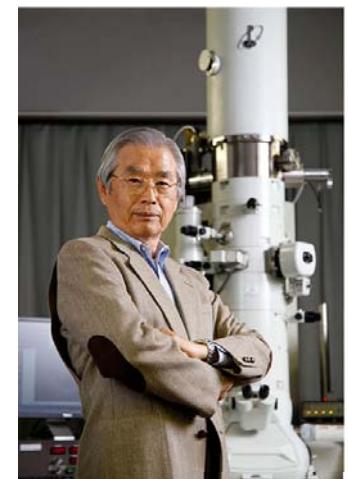
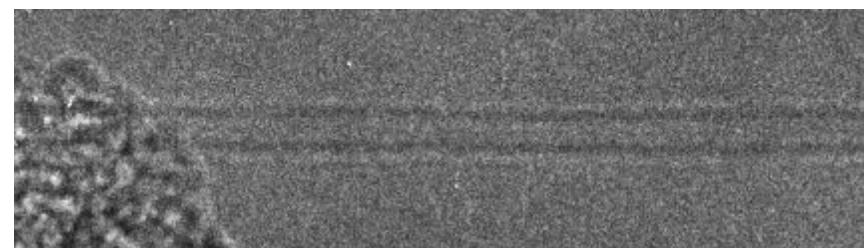
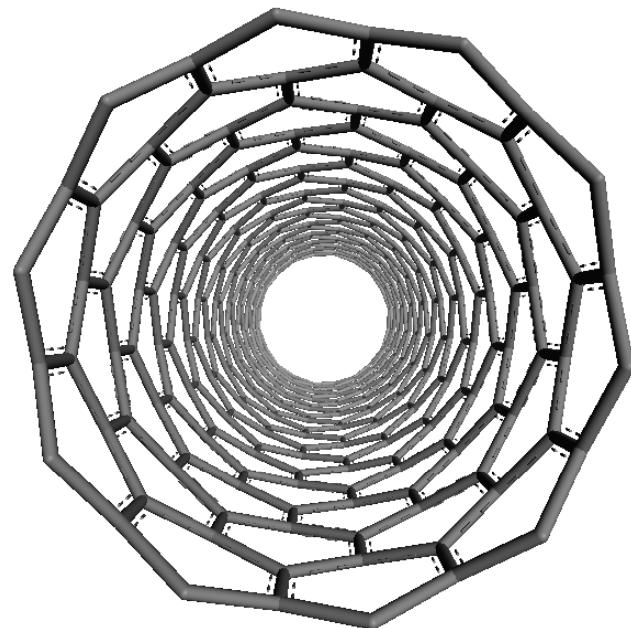


Fig. 3. Schematic illustration of rollers



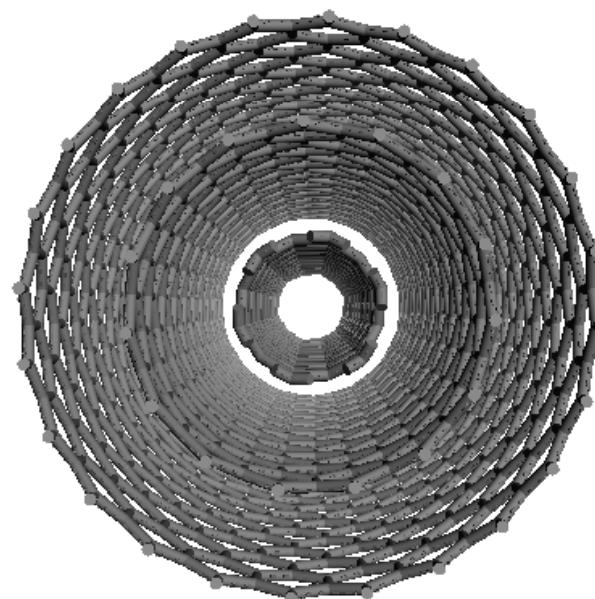
Structure of CNT

Single-Walled CNT
(SWCNT)



(10,0)

Multi-Walled CNT
(MWCNT)

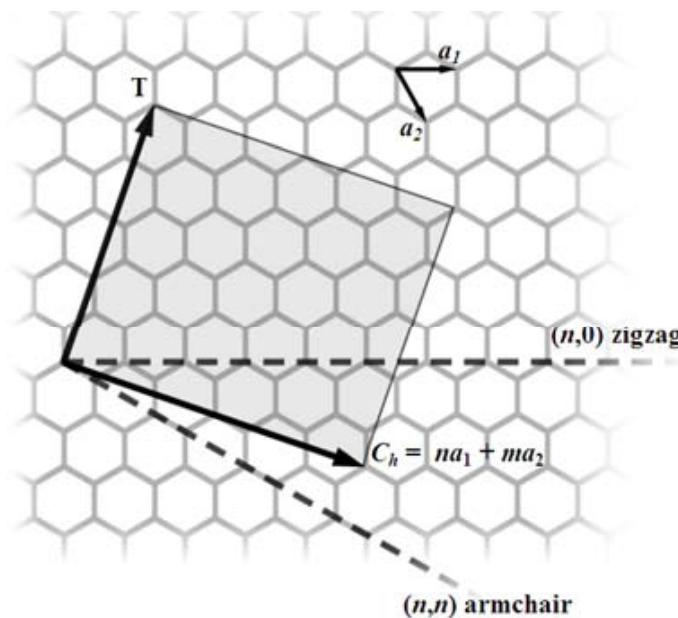


(6,0)+(15,0)+(24,0)

The interlayer distance in multi-walled nanotubes is close to the distance between graphene layers in graphite, approximately **3.4 Å**.

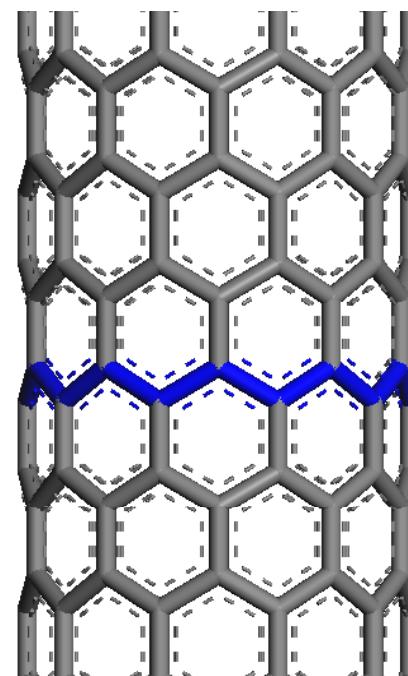
Structure of CNT

- Similar to GNRs, we should take the chirality of CNTs into account

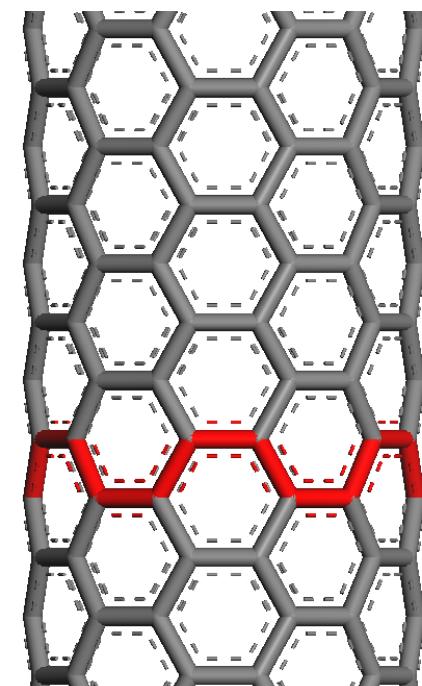


The diameter of $[n,m]$ CNT:

$$d_{CNT} = \frac{a}{\pi} \sqrt{n^2 + nm + m^2}$$



(10,0) Z-CNT

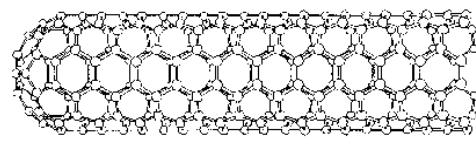
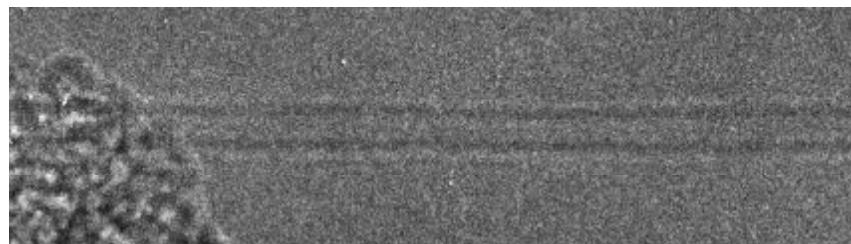


(6,6) A-CNT

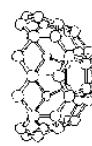
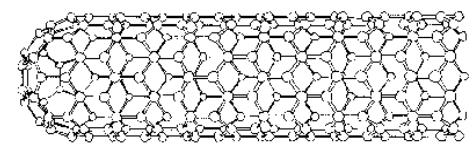
Physical Properties of CNT

- Semiconductor or metal. (depending on the chiral angle)
- Excellent mechanical properties
- Excellent electrical transport (ballistic transport)
- Excellent thermal conductivity
- Difficulty– Control the chirality

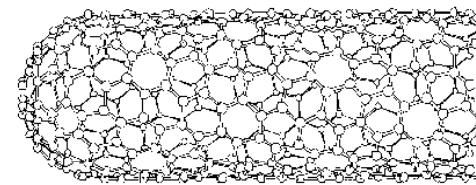
1-D nanomaterials (Carbon nanotube)



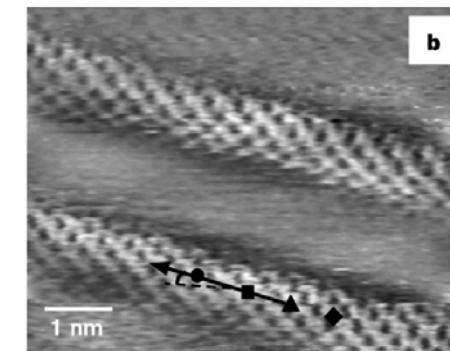
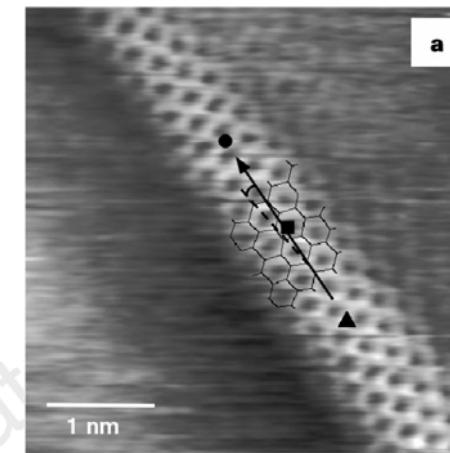
Armchair (n,n)



Zig-zag (n,0)

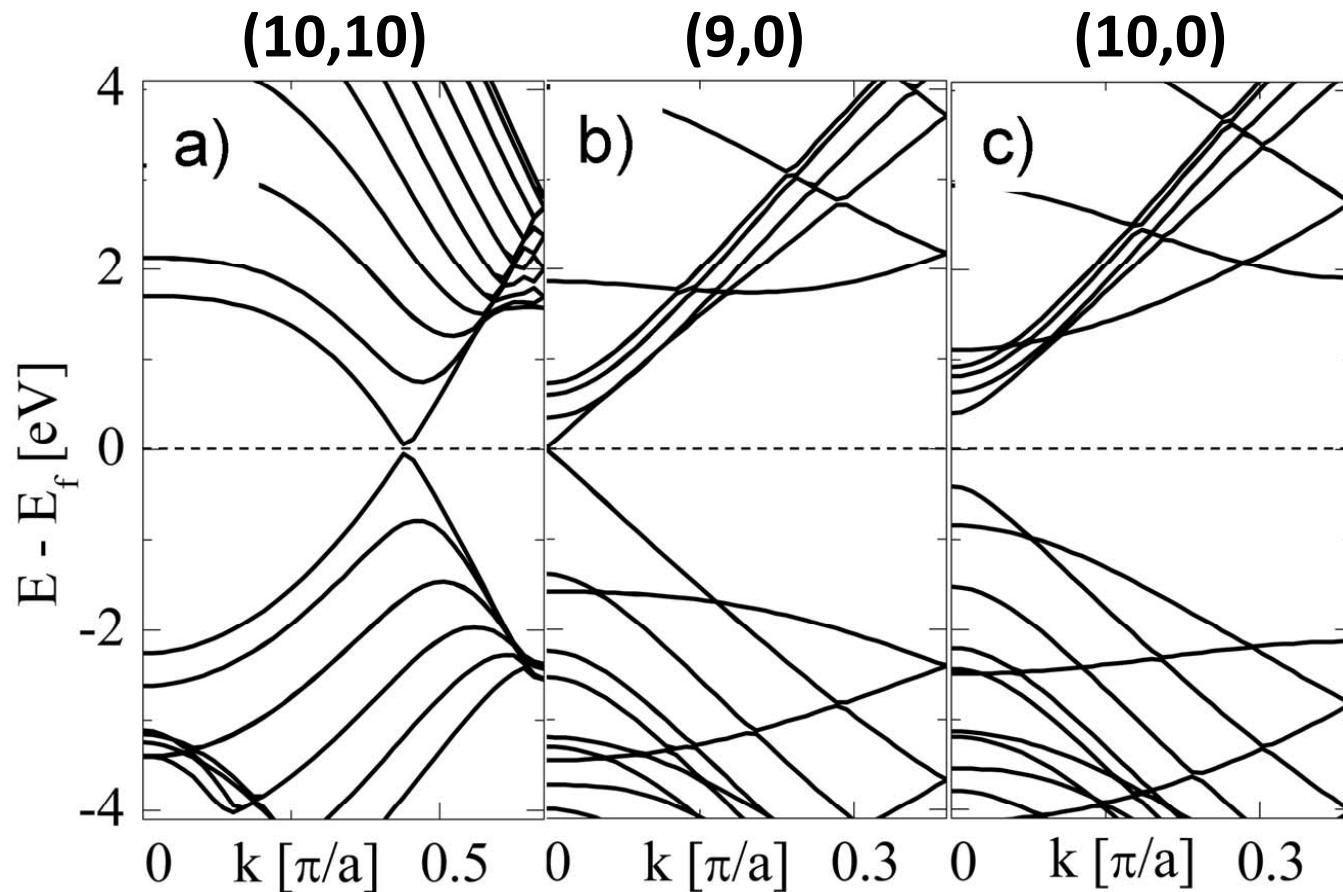


Chiral (n,m)



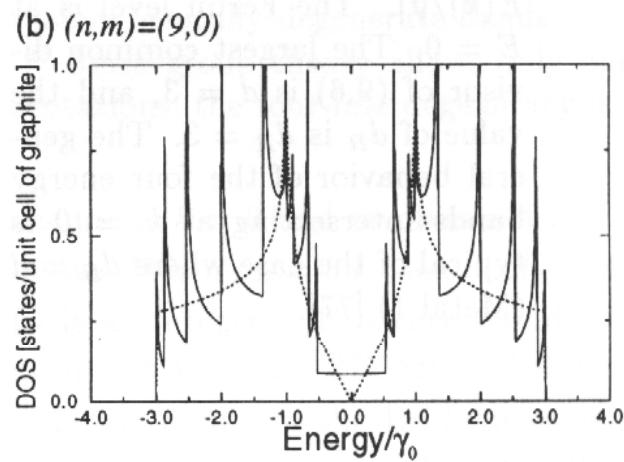
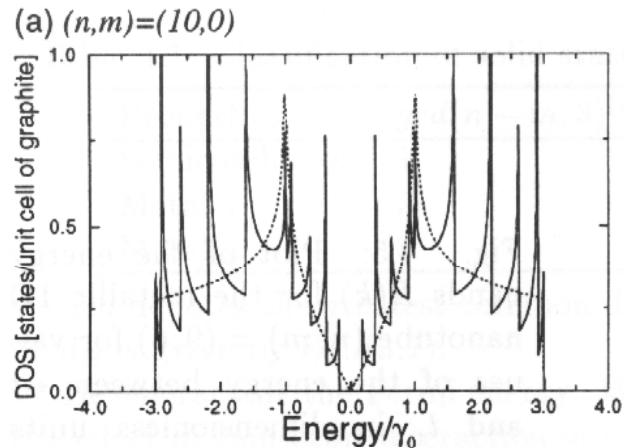
Discovered by Iijima (1991)

Band Structures of CNT

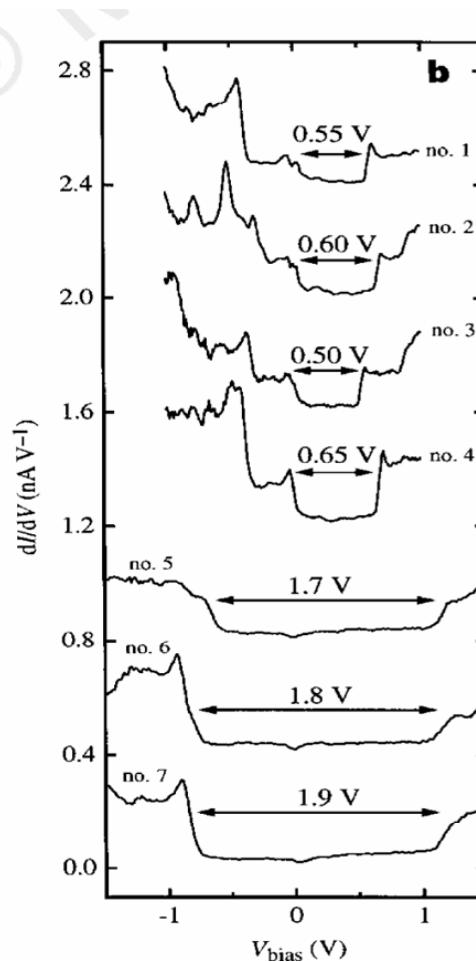


Band structures of (a) (10,10) (b) (9,0) (c) (10,0) CNTs

Electronic Properties of CNT



Tight-binding
calculations



Experiments

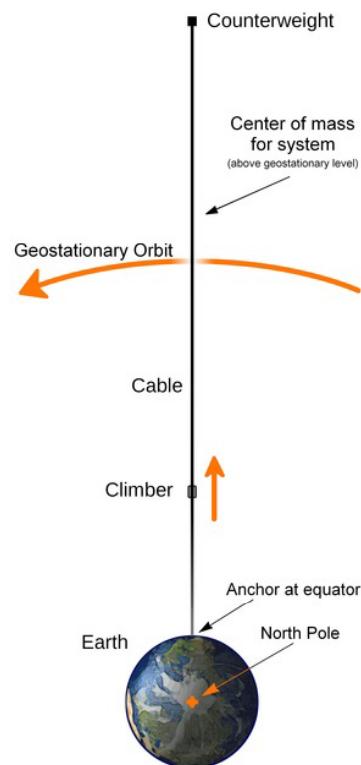
Energy gap changes
with *chirality* and
diameter

Nature 391, 62, 1998

Structural Materials

- Due to the outstanding mechanical properties and light-weighted, CNT is a potential material for some special usages

Space elevator



Bulletproof cloth, shield, glass



Bicycle components



Mechanical Properties

- CNT is one of the strongest materials in nature
 - Very strong in the axial direction

| | Young's modulus (GPa) | Tensile strength (GPa) | Elongation at break (%) |
|---------|-----------------------|------------------------|-------------------------|
| SWCNT | >1000 | 13-53 | 16 |
| Steel | 186-214 | 0.38-1.55 | 15-50 |
| Diamond | 1220 | 2.8 | - |

- Very hard; even harder than the diamond

| | Bulk modulus (GPa) |
|---------|--------------------|
| SWCNT | 462-546 |
| Steel | 160 |
| Diamond | 442 |

Belluci, S. et al., *Phys. Status Solidi C* **2** (1), 34 (2005)
Sinnott, S.B. et al., *Crit. Rev. Solid State* **26** (3), 145 (2001)

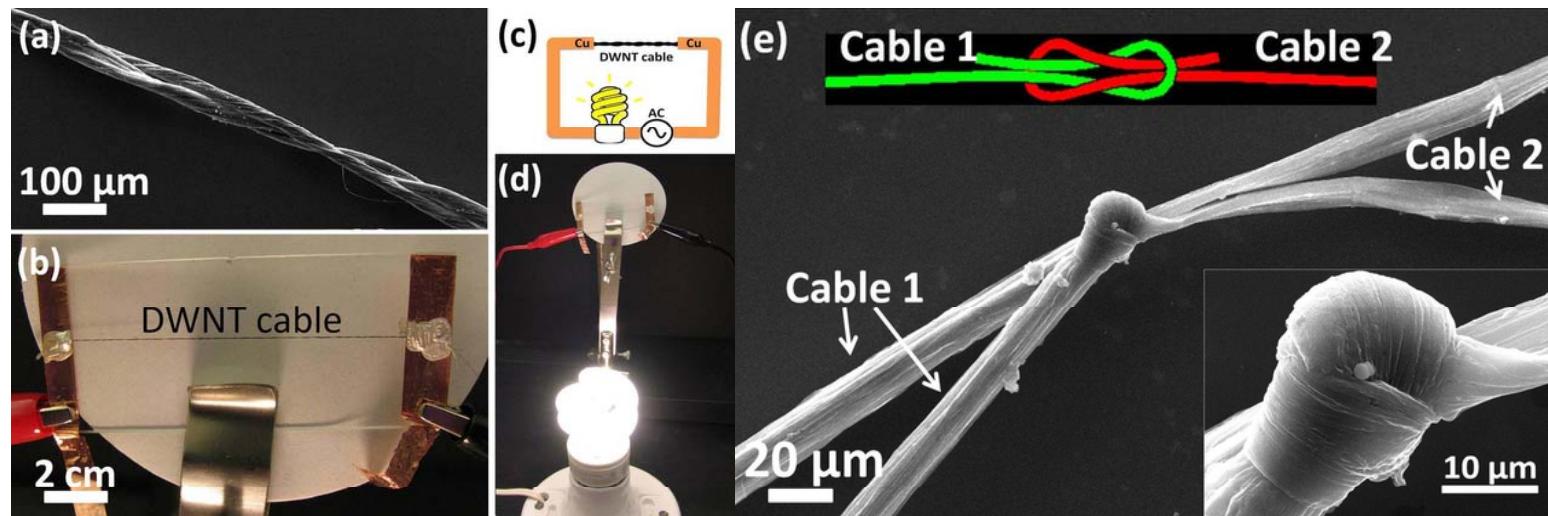
Electronic Properties

- CNTs are semiconducting or metallic, depending on the chirality
- High mobility ($100,000 \text{ cm}^2/(\text{V}\cdot\text{s})$ at room temperature)
- High electric current density of metallic CNT ($4 \times 10^9 \text{ A/cm}^2$, which is over 1000 times of Cu)
- Superconductivity (no clear evidence, under debating)

Tang, Z. K. et al., Science 292 (5526), 2462 (2001)

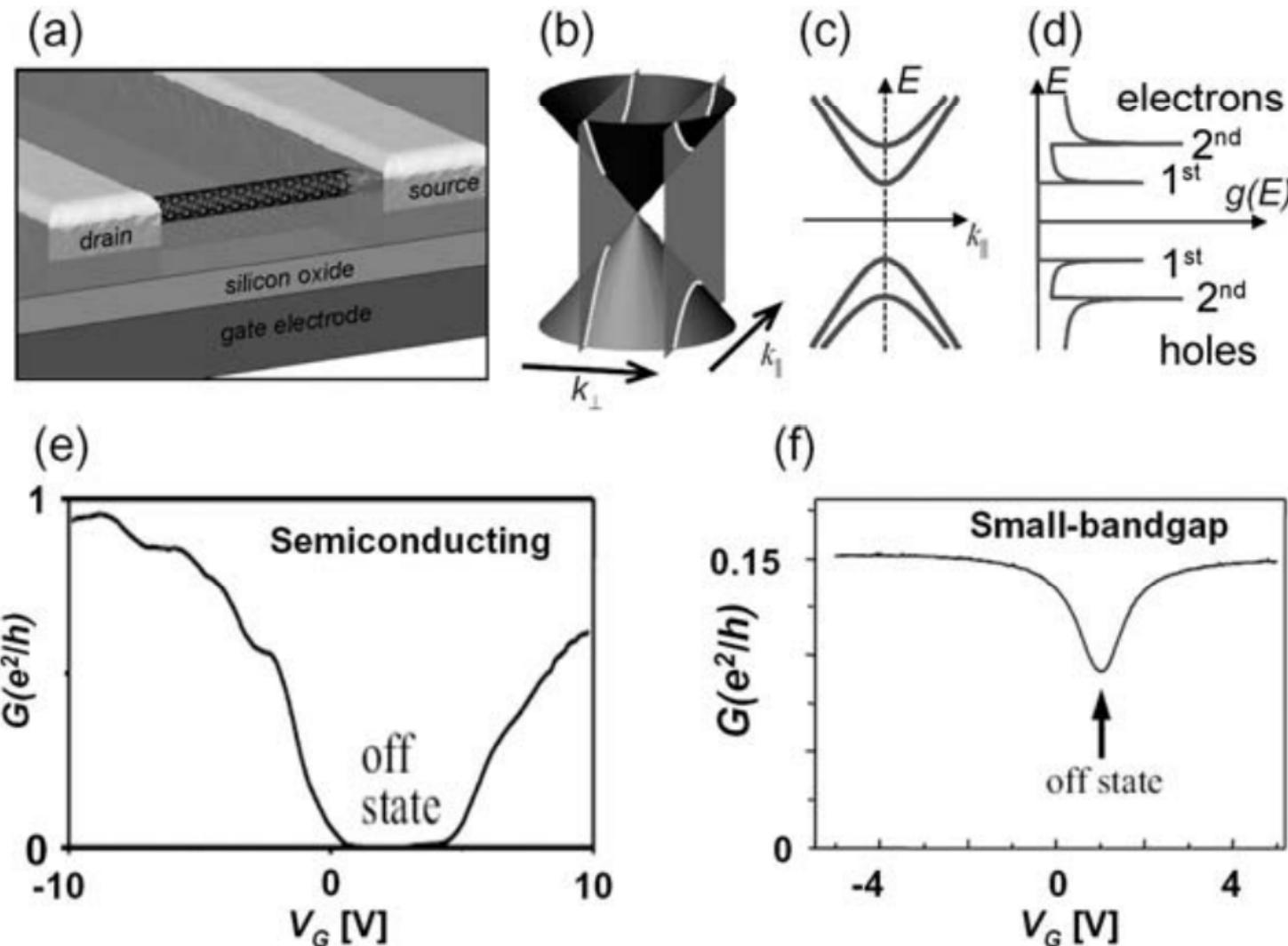
Interconnects

- Similar to GNRs, CNTs are potential materials to replace metal (such that Cu, Au) as non-metal high-conductivity interconnects in the integrated circuits



Transport of CNT

Nanotube field-effect transistor

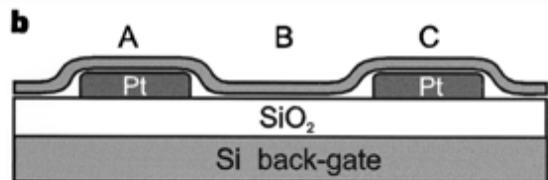


Ambipolar Electrical Transport in
Semiconducting Single-Wall Carbon Nanotubes

CNT Transistor

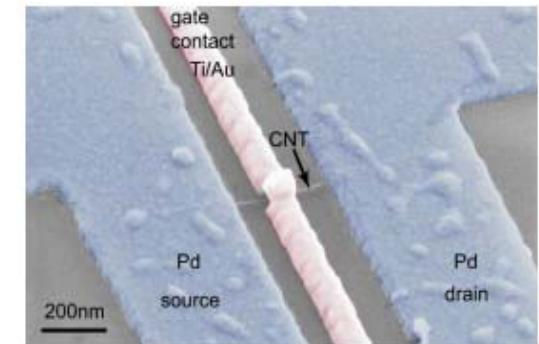
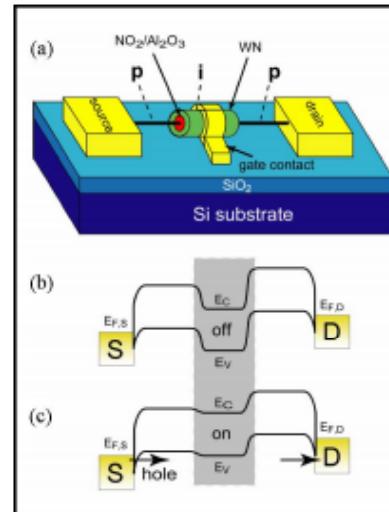
- Various type of CNT transistors had been developed

Back-gated



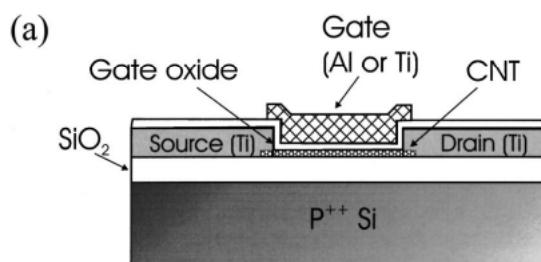
Sander J. Tans, et. al., *Nature* 1998, **393**, 49-52

Wrap-around gate



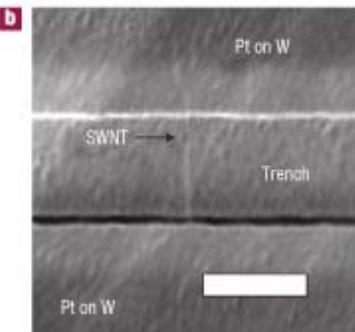
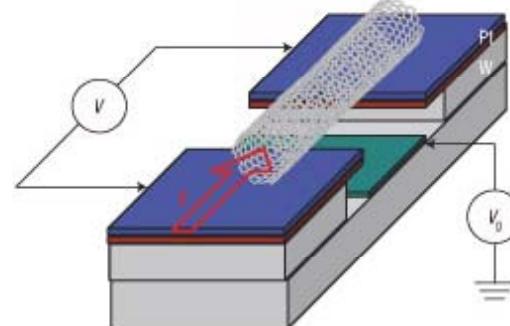
Chen, Z., et. al., *IEEE ELECTRON DEVICE LETTERS*, VOL. 29, NO. 2, 0741.

Top-gated



Wind, S. J. et. al., 2002 *Applied Physics Letters* **80** (20): 3817

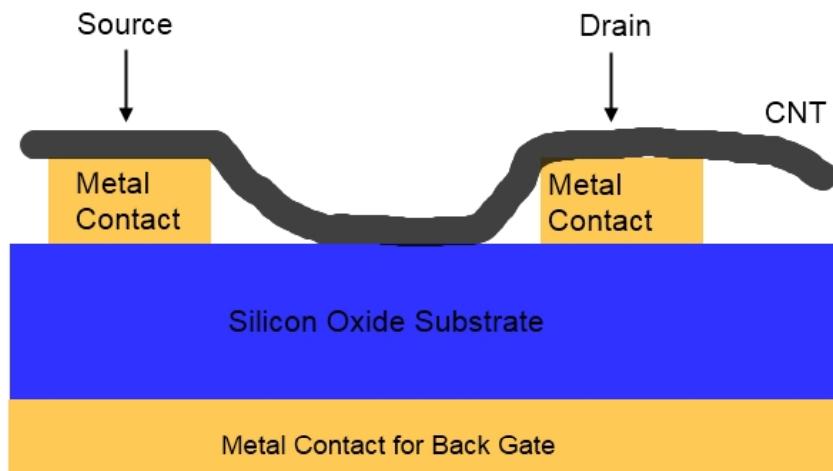
Suspended



Cao, J., et. al. *Nature Materials* **4**, 745 - 749 (2005)

Designs of CNT Transistors

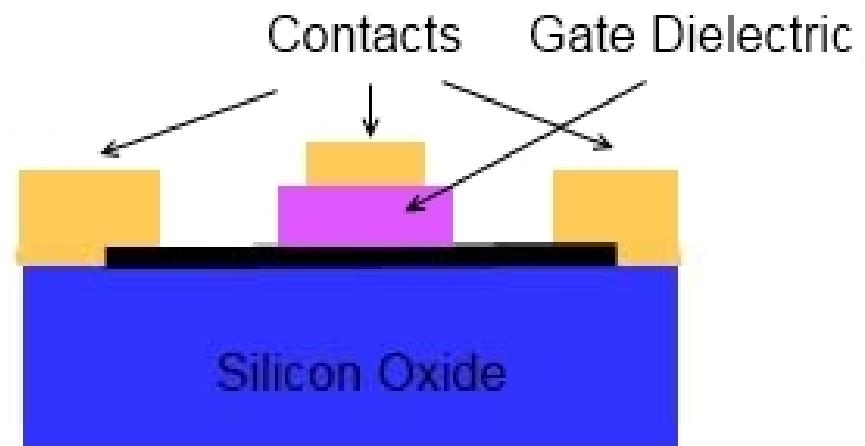
Back-gated



First CNT FET be invented. The process is easy, but there are many drawbacks

1. Contact resistance exists between metal and CNT
2. Hard to switch off using low voltage
3. Poor contact between SiO_2 and CNT

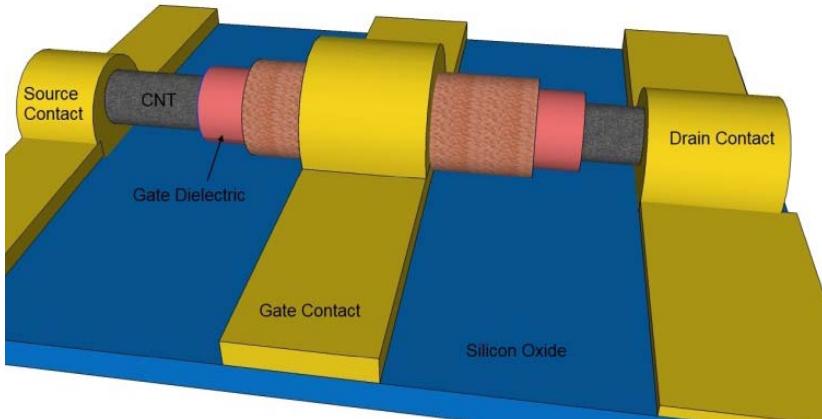
Top-gated



Advanced designs from back-gate FET. The thinner gate dielectric lower the switch-off voltage, which is the main advantage of the top-gate FET.

Designs of CNT Transistors

Wrap-around gate



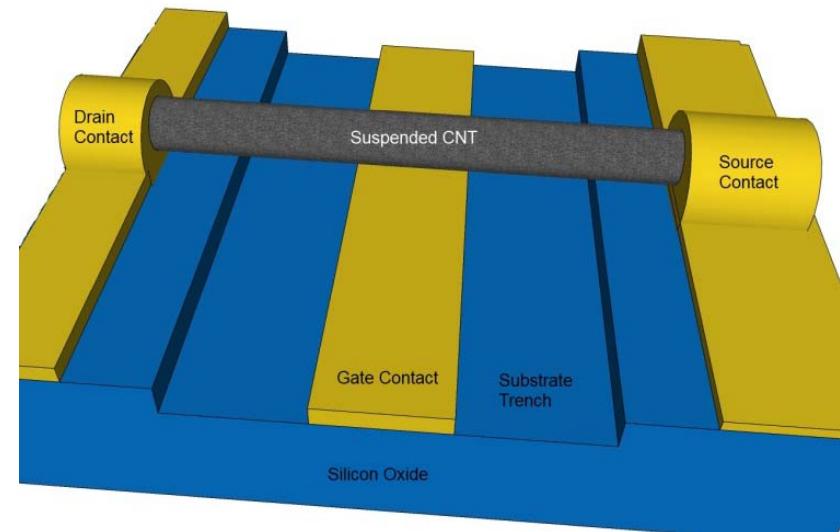
Developed in 2008. This design

1. Improve the on/off ratio and the performance
2. Reduce the leakage current

Z. Chen et al., *EDL* **29** (2), 183 (2008)

J. Cao et al., *Nature Materials* **4**, 745 (2005)

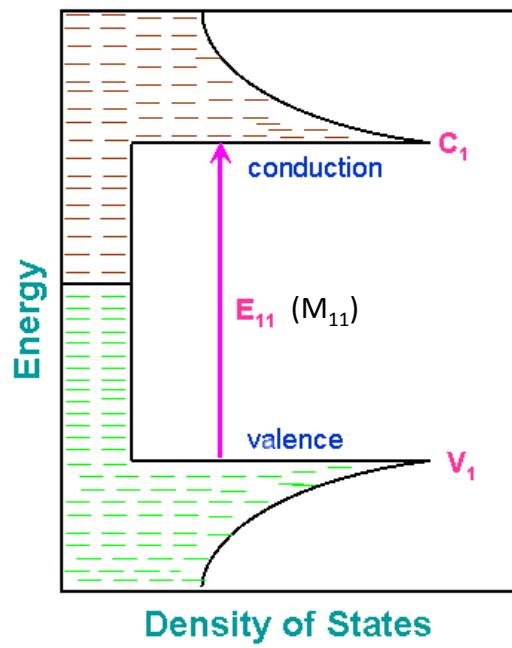
Suspended



Developed in 2005. This design reduces the scattering at the interface of the CNT-substrate.

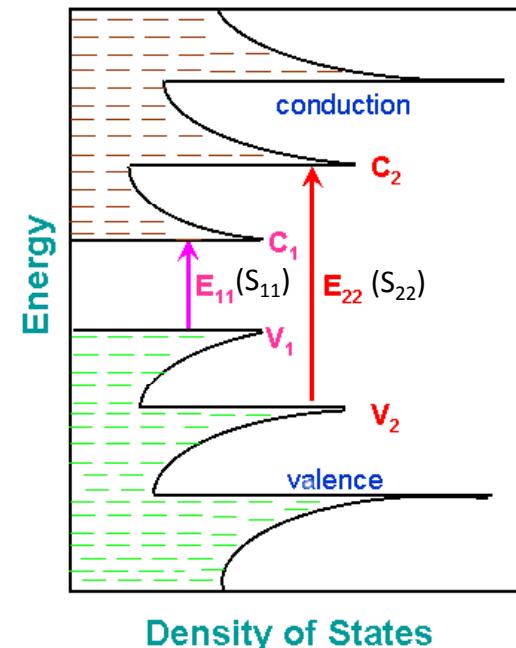
However, materials of devices are limited. It is only for studying the properties of clean CNT and cannot be commercialized.

Optical Properties of CNT



Metallic SWNT

$v_1 \rightarrow c_1$ corresponds to the “first van Hove” optical transition



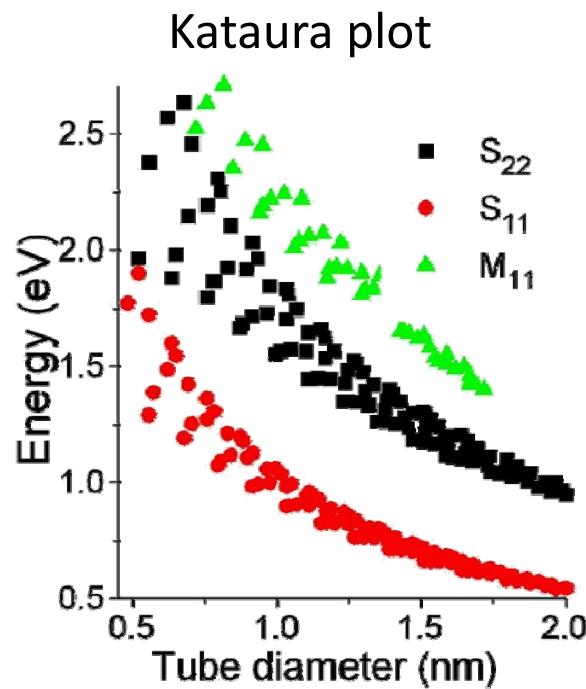
Semiconducting SWNT

$v_2 \rightarrow c_2$ corresponds to the “second van Hove” optical transition

Van Hove singularity: The discontinuity of DOS of 1D materials

Optical Properties

- Because of the sharp transitions, we can selectively excite (n,m) CNT; namely, we can detect optical signals in order to identify (n,m) CNT

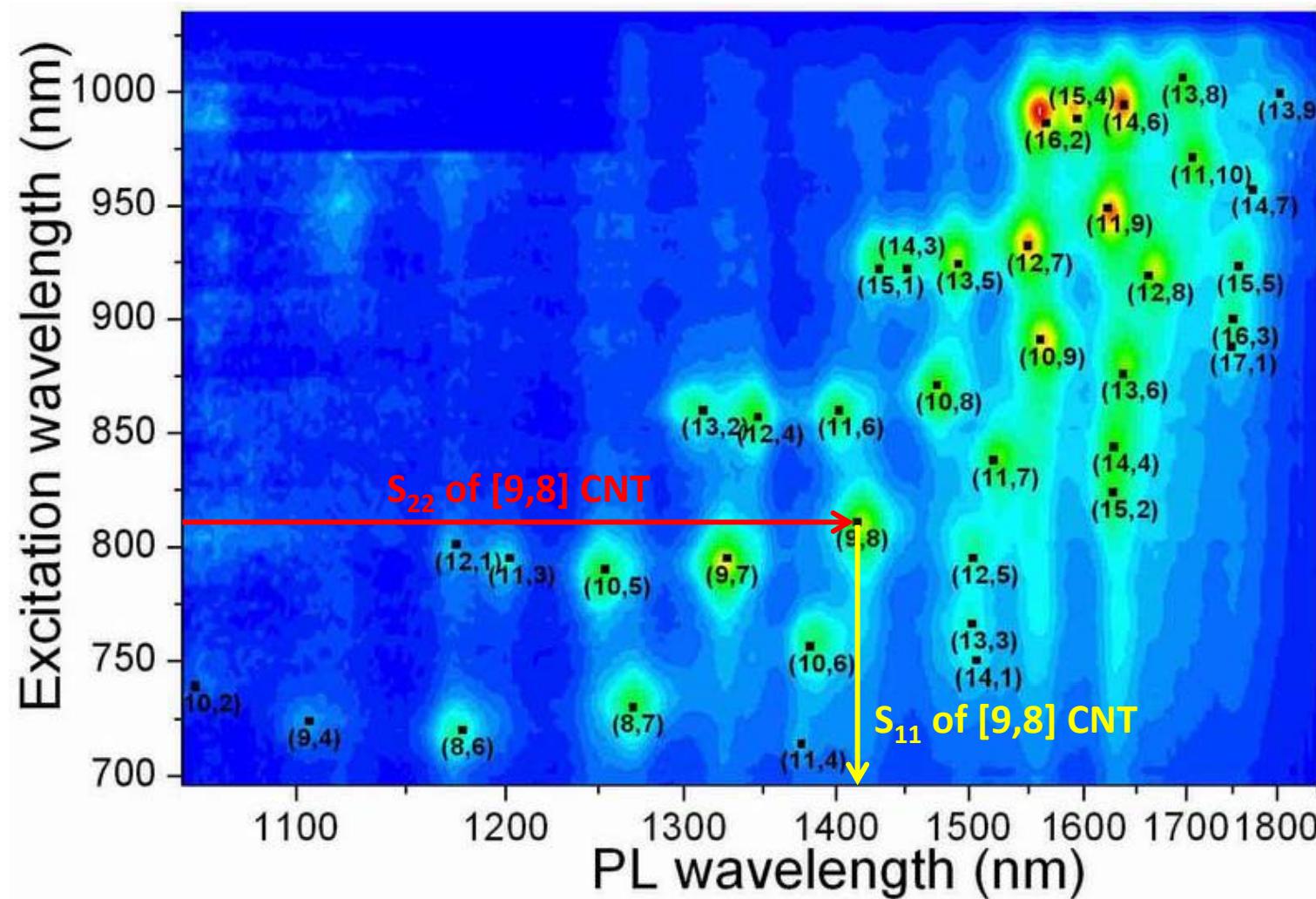


Photoluminescence of CNT

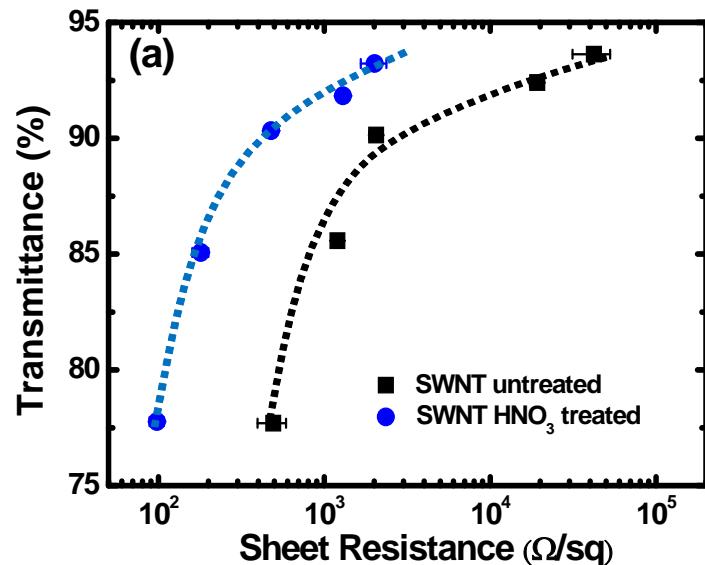
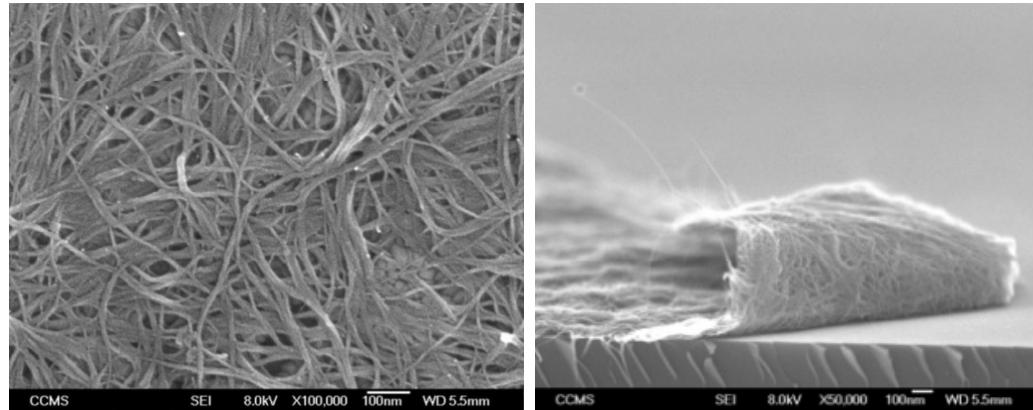
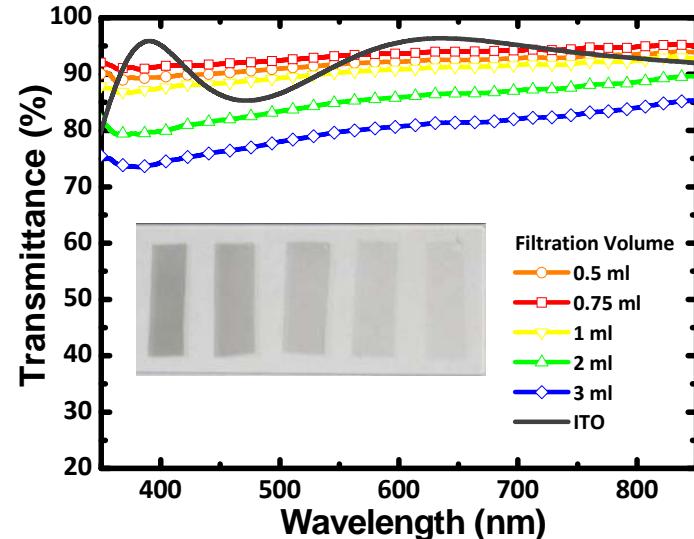
- Process of PL
 1. Excite S_{22} (C_2-V_2) and create the exciton pair
 2. Relax to C_1 and V_1 respectively
 3. Recombine and emit light with S_{11} (C_1-V_1) energy
- An important tools for characterizing semiconducting CNT

R. B. Weisman and S. M. Bachilo, *Nano Lett.* **3**, 1235 (2003)

PL Mapping of CNT

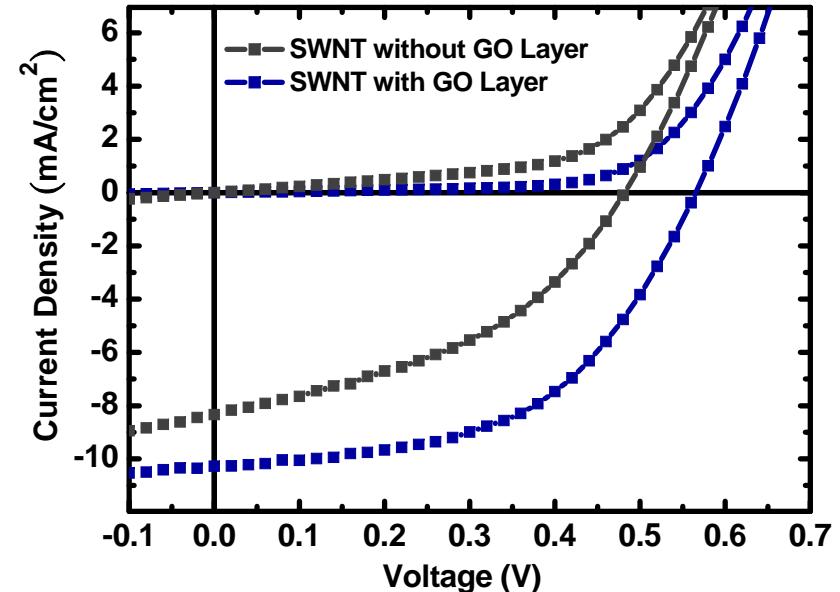
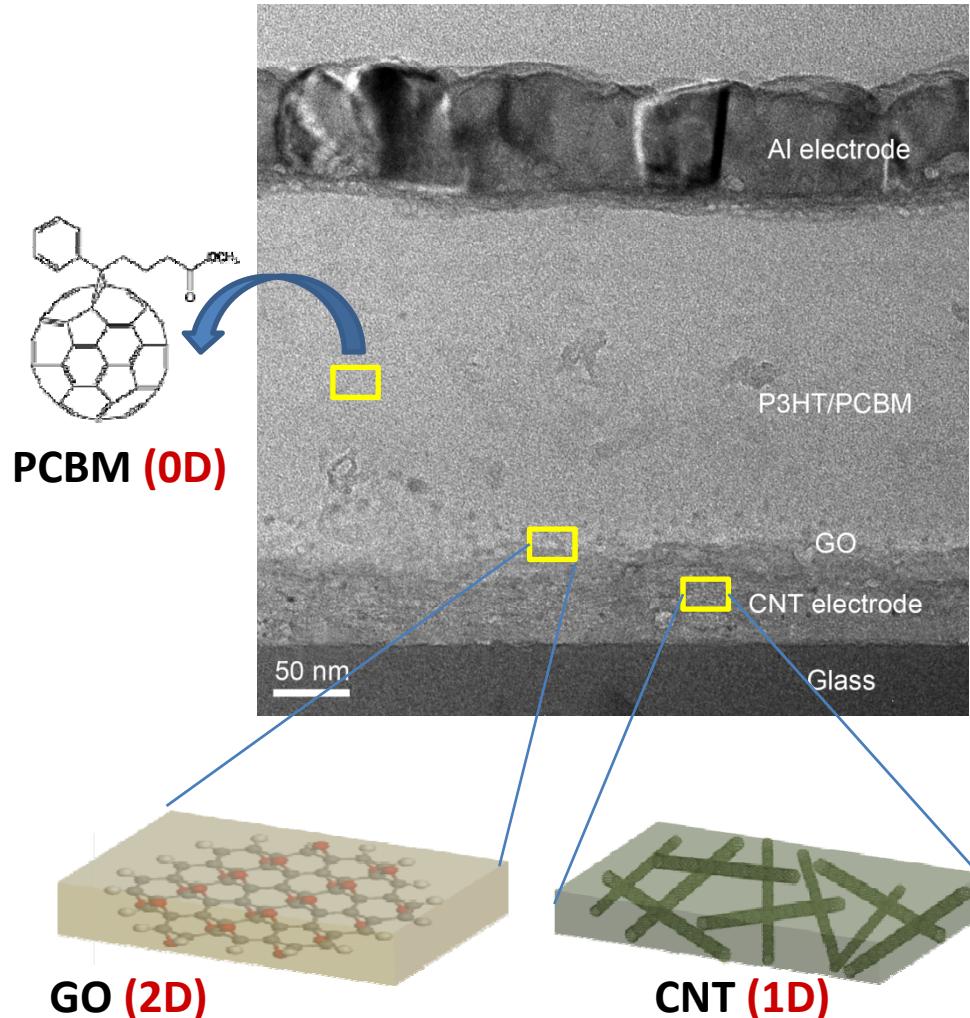


Applications: Optical and Electrical Properties of SWNT Electrode



- SWNT electrode has high flexibility and excellent mechanical strength
- Sheet resistance of SWNT electrode can be further reduced by HNO_3 treatment
- Sheet resistance $\sim 100 \Omega/\text{sq}$ with 75% transparency @ 550nm

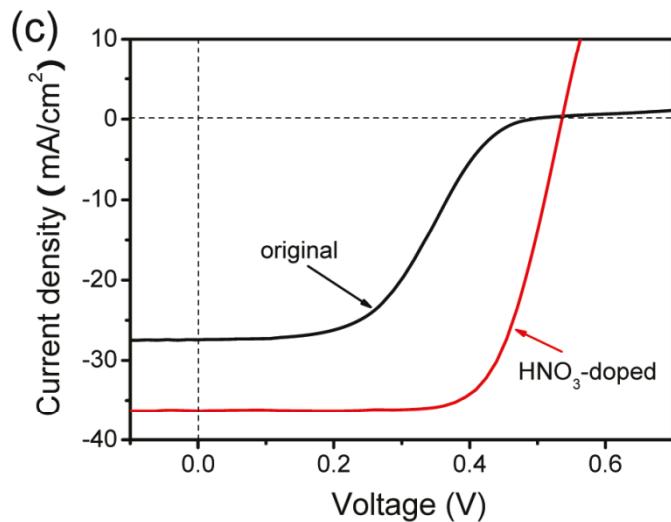
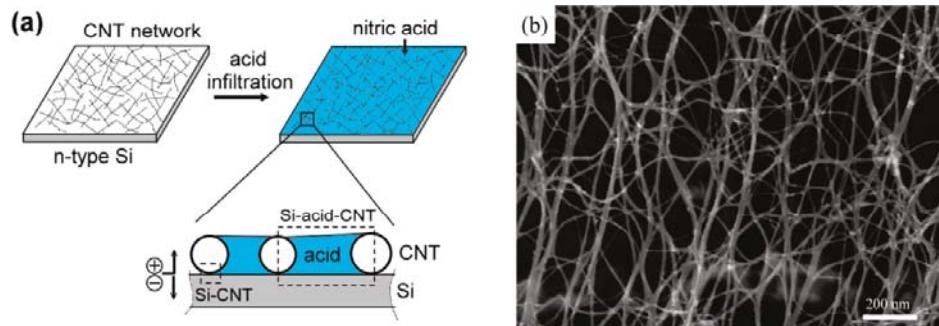
“Cocktail” nanocarbon polymer solar cell !



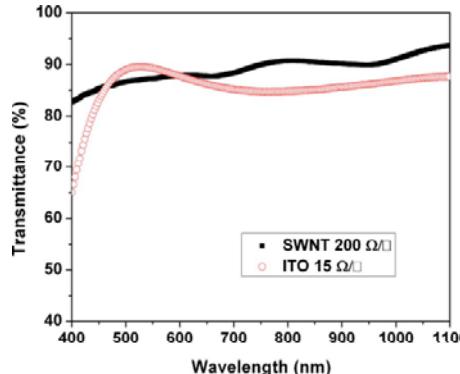
| SWNT | V _{oc} (V) | J _{sc} (mA) | Fill Factor | PCE (%) |
|---------------|---------------------|----------------------|-------------|---------|
| Without GO | 0.48 | 8.33 | 0.42 | 1.67% |
| With GO | 0.57 | 10.3 | 0.53 | 3.1% |
| ITO/PEDOT:PSS | 0.57 | 11.0 | 0.57 | 3.57% |

Carbon Nanotube on photovoltaic and LED applications

Silicon-CNT heterojunction solar cell

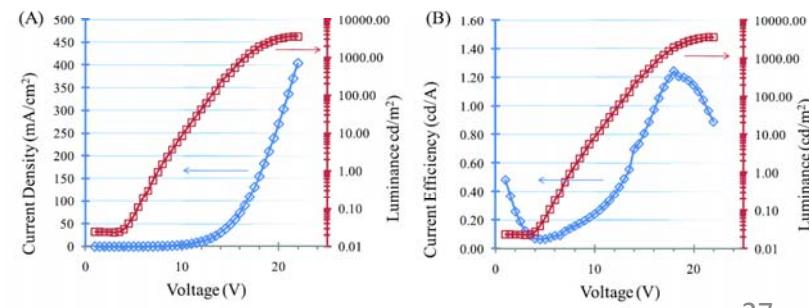
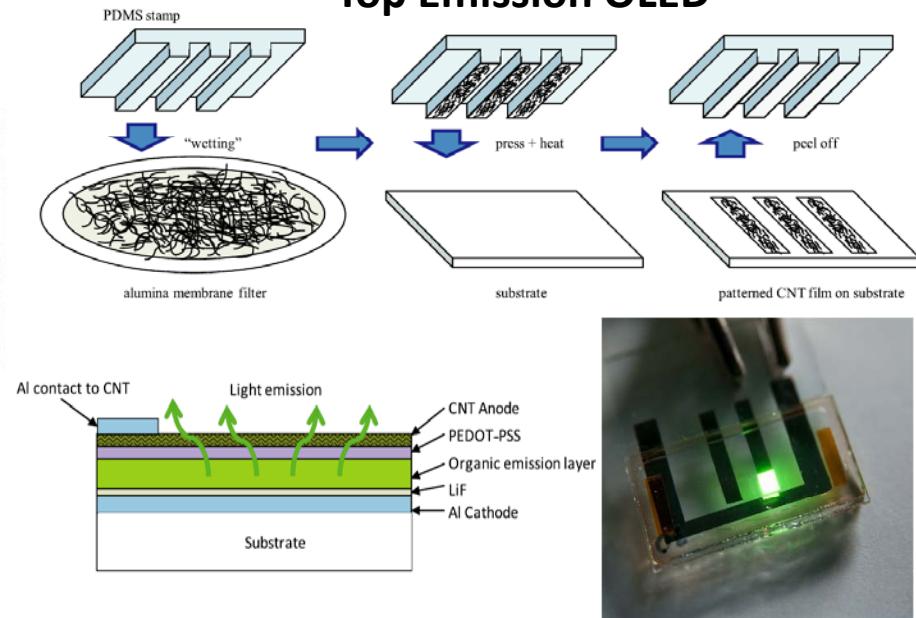


Jia, Y. et. al., *Nano Lett*, 2011, 11, 1901.



- ❖ High transparency on visible and NIR range
- ❖ Good conductivity
- ❖ Flexible

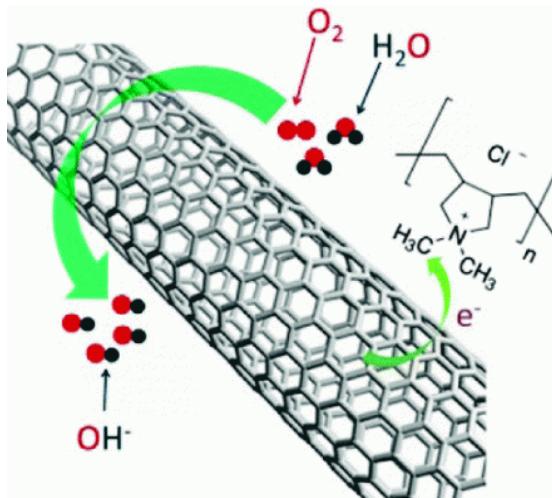
Top Emission OLED



Chien, Y. M. et. al, *Nanotechnology*, 2010, 21, 134020

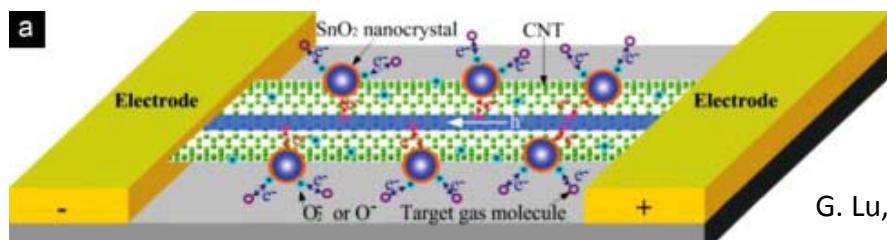
Other Applications

- Fuel cells
 - Store H₂ in CNT



Wang, S. Et. al., J. Am. Chem. Soc., 2011, 133 (14), 5182

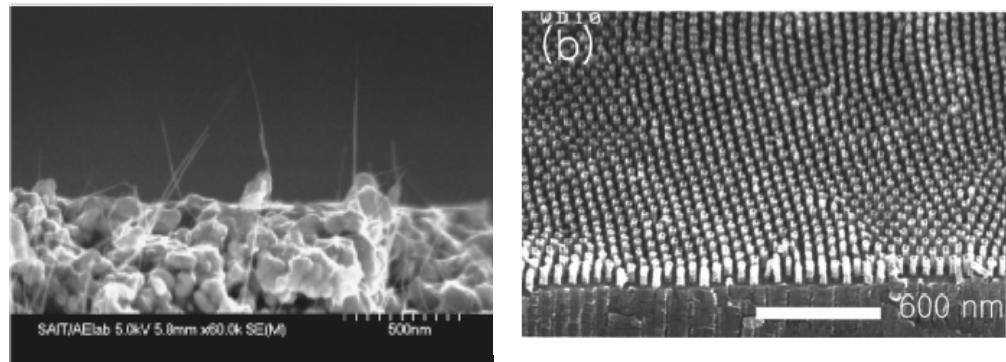
- Gas detector
 - Molecules adsorb on the channel of CNT FET, modifying the electrical properties of CNT



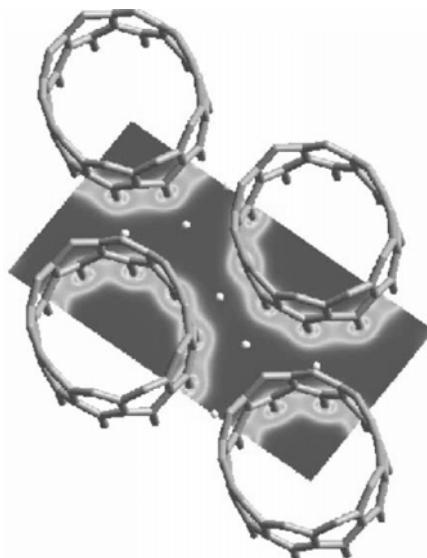
G. Lu, L.E. Ocola J. Chen, Adv. Mater, 2009, 21, 2487

Applications of Carbon Nanotube

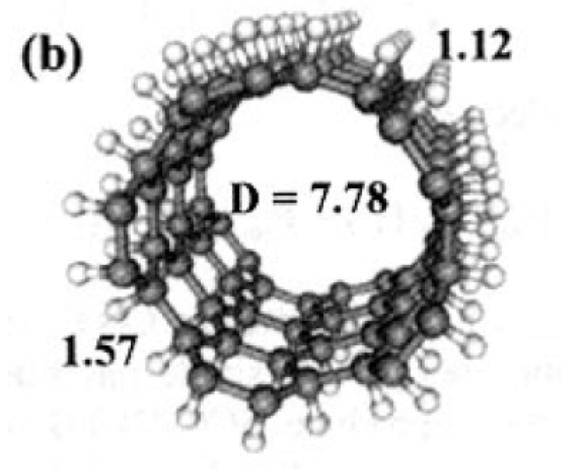
Field Emitter



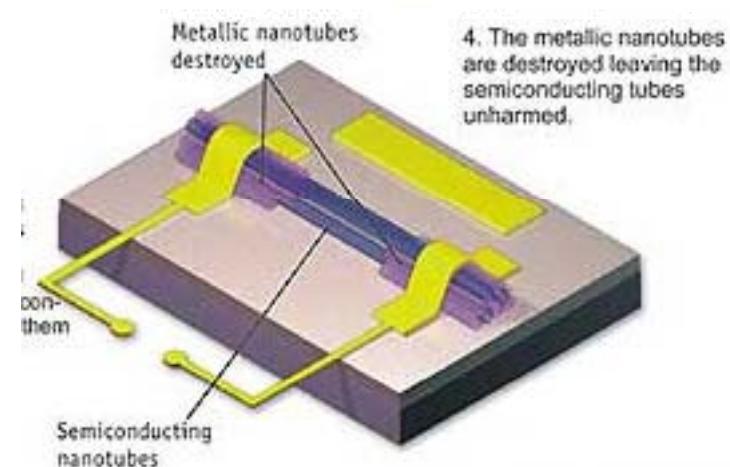
Li-ion battery anode



Hydrogen Storage

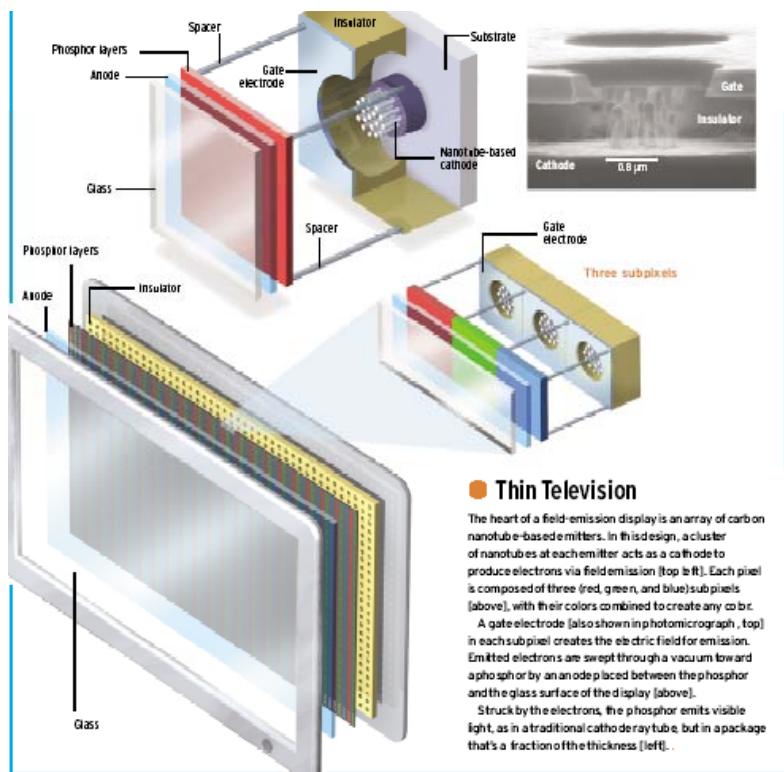


Logic gate by IBM



Applications of CNTs in electronic devices

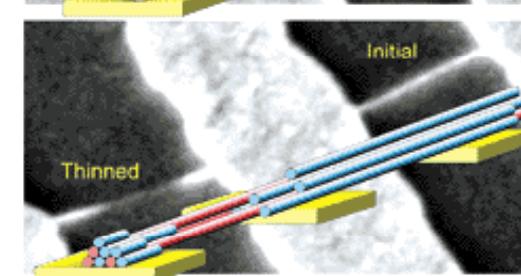
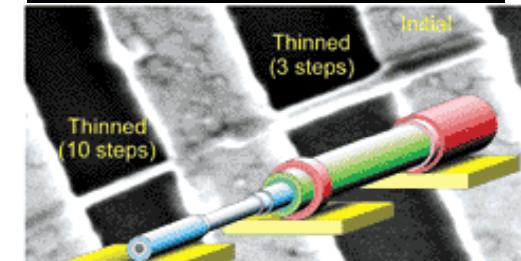
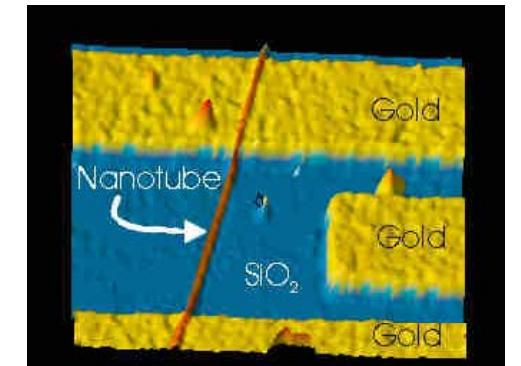
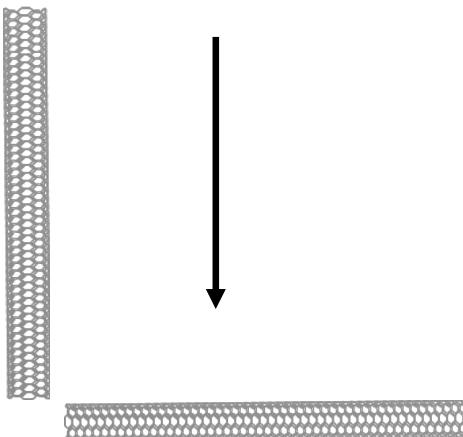
CNT field emission display



From IEEE spectrum 2003

CNT Field effect transistor

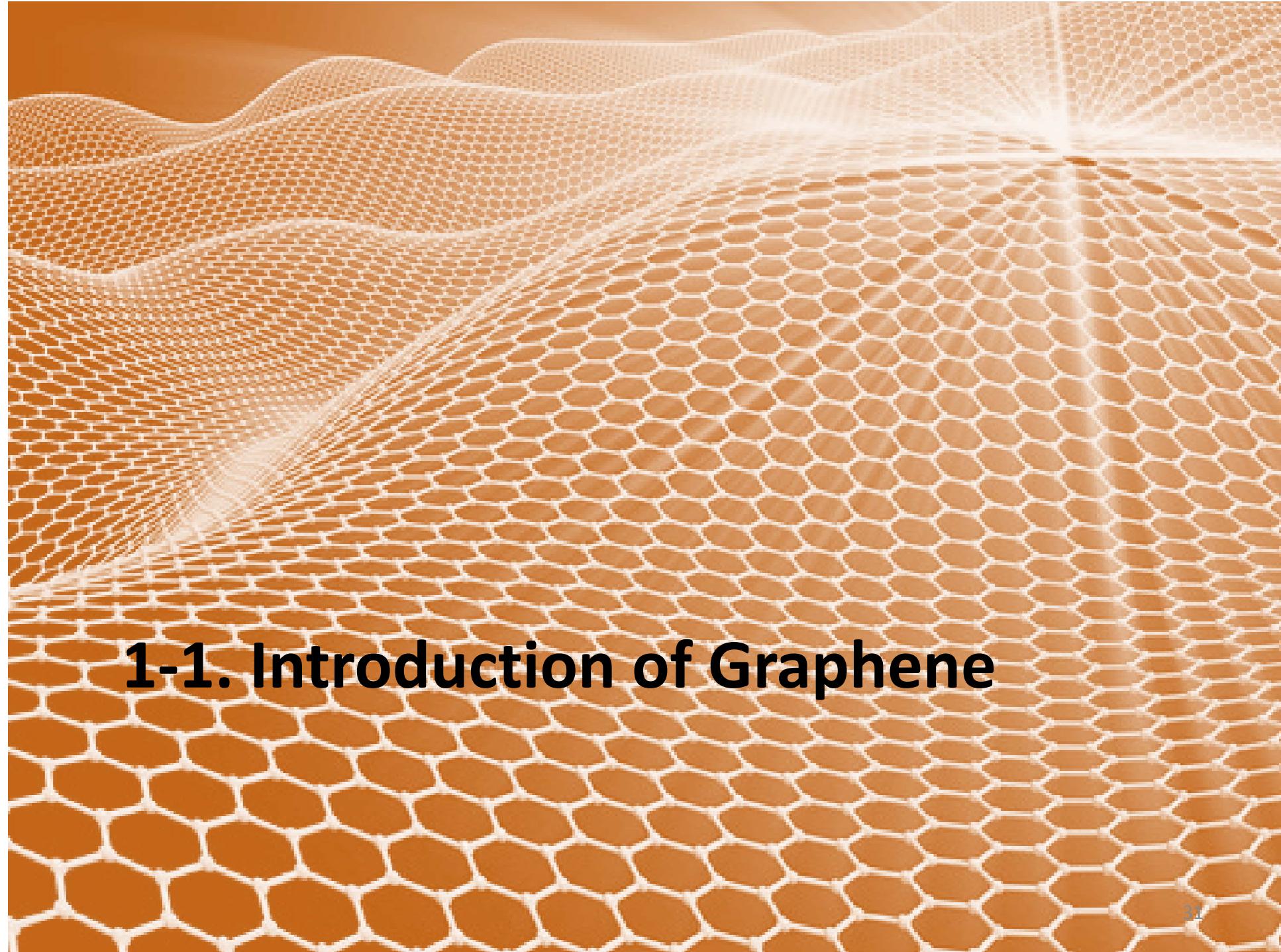
E field



Small dimension

- High local field
- Electronic structure changes

From IBM

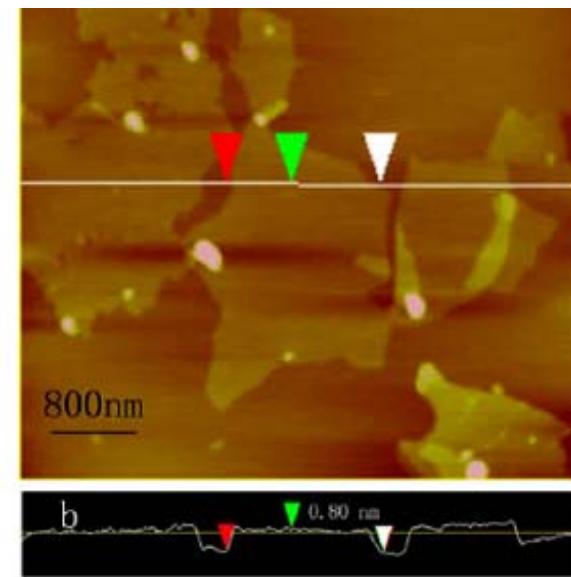
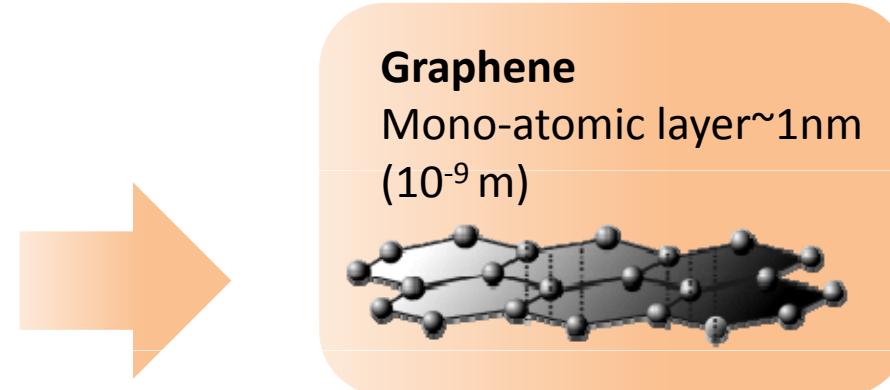
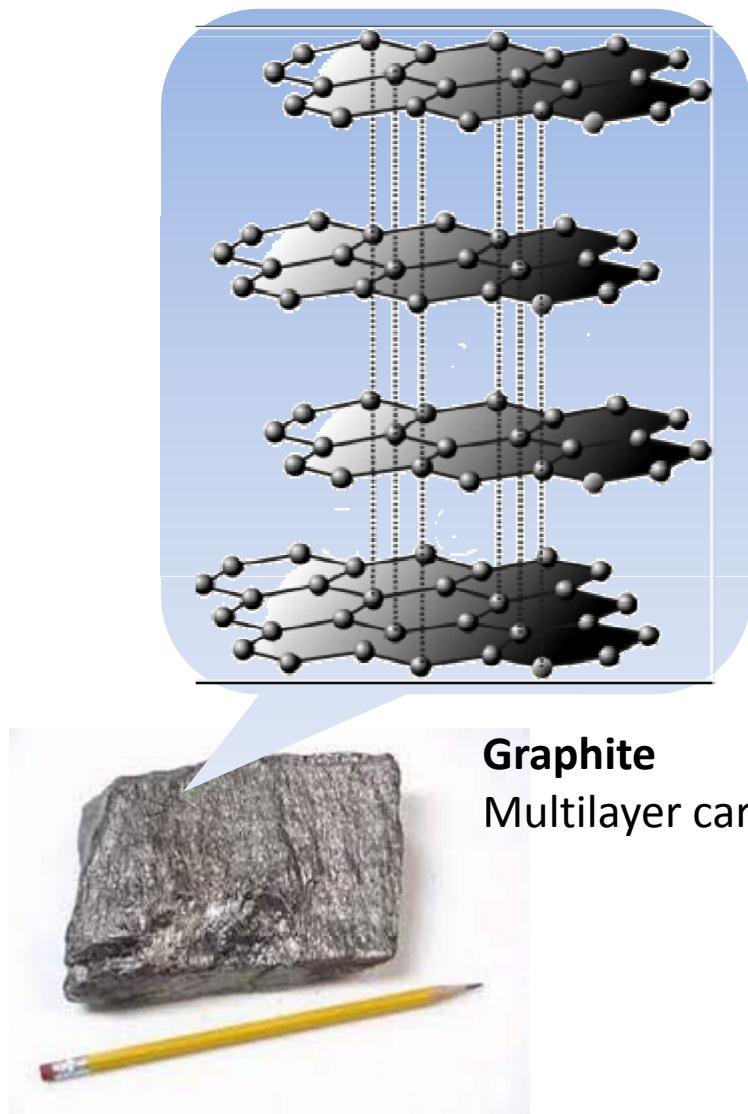


1-1. Introduction of Graphene

Graphene History

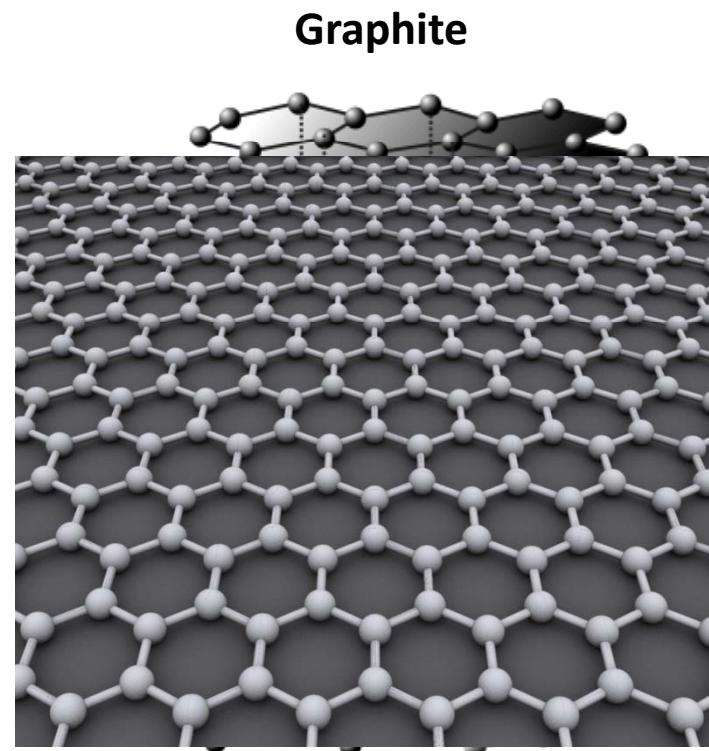
- Early: Theoretical description
- 1962: Named by Hanns-Peter Boehm (Graphite + -ene)
- 2004: Single-atom-thick, free-standing graphene is extracted (by Andre Geim and Konstantin Novoselov, Manchester University, U.K.)
- 2005: Anomalous quantum Hall effect was observed
- 2010: Nobel prize in Physics for Andre Geim and Konstantin Novoselov
- Now: Stimulate wide researches and be applied to various fields

From graphite to graphene



Graphene

- What is graphene?
 - A single layer of graphite
 - 2D material
- Structure
 - One-atom-thick planar sheet
 - sp^2 -bonded carbon atoms
 - Honeycomb lattice



2010 The Nobel Prize in Physics



Photo: Sergeom, Wikimedia Commons

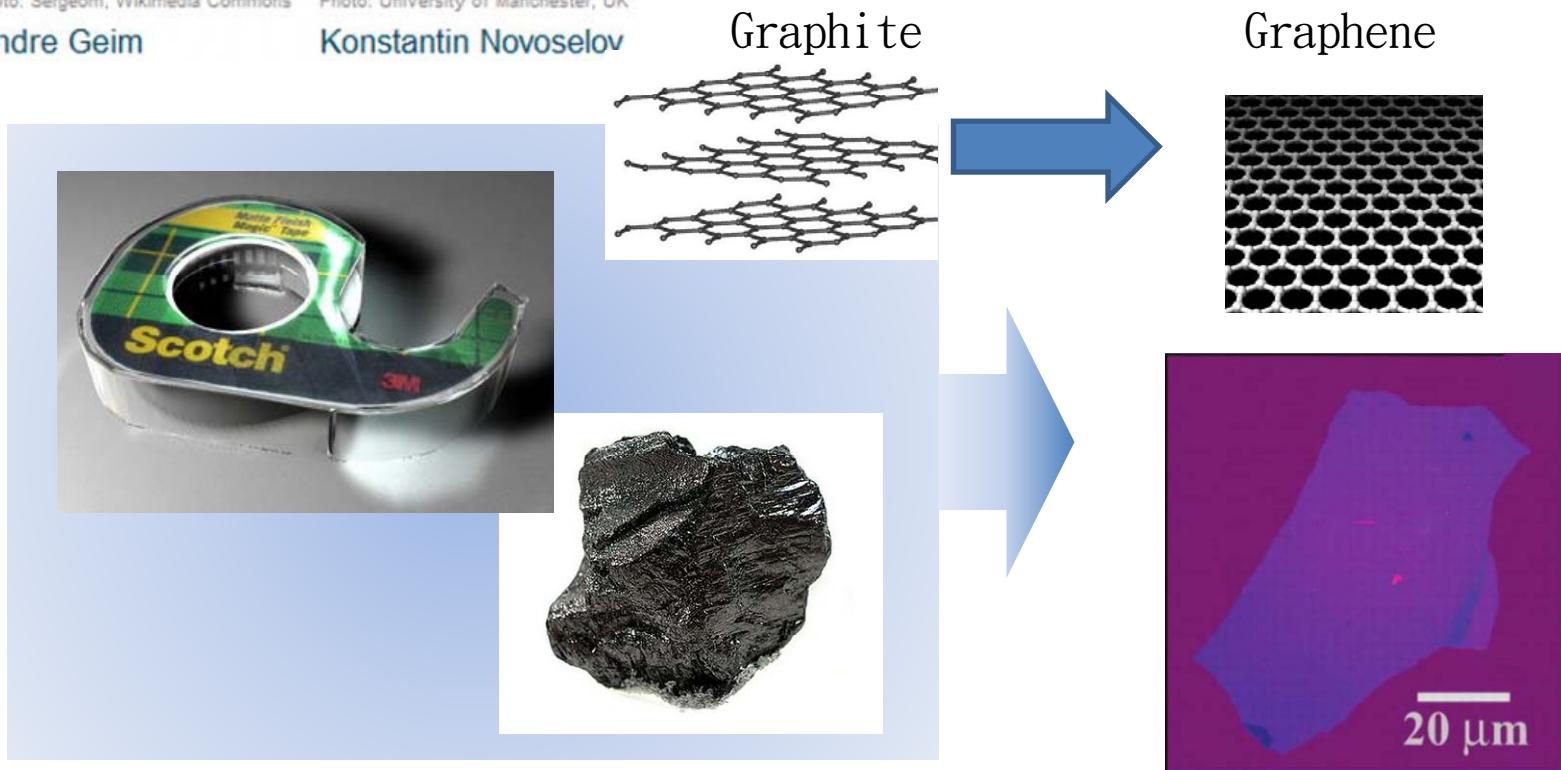
Andre Geim



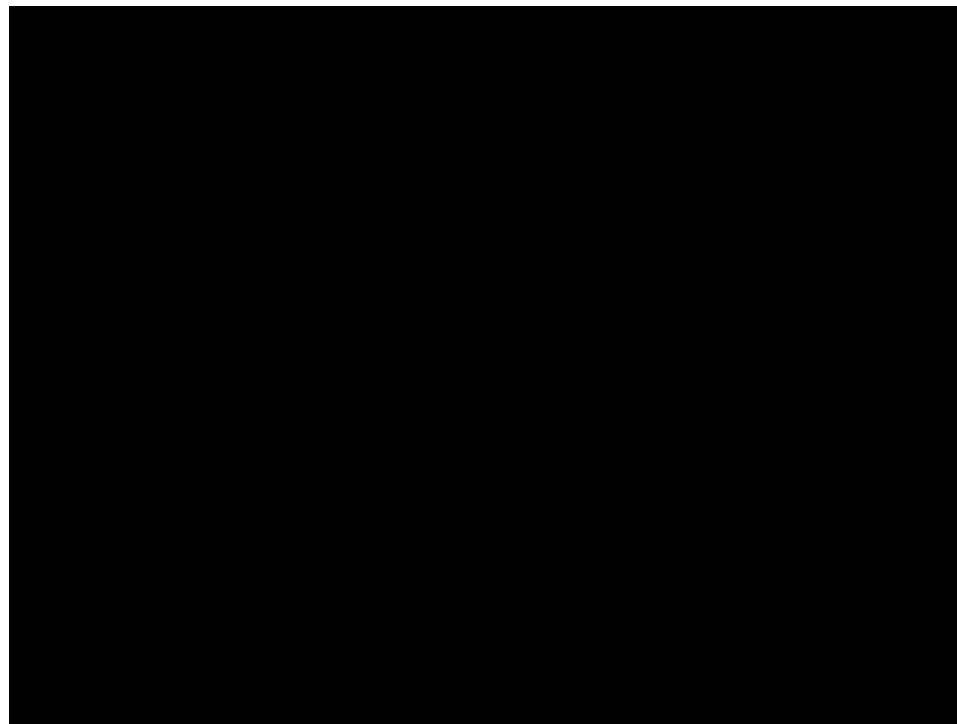
Photo: University of Manchester, UK

Konstantin Novoselov

Prof. Andre Geim and Konstantin Novoselov at the U. Manchester for groundbreaking experiments regarding the 2-D material graphene

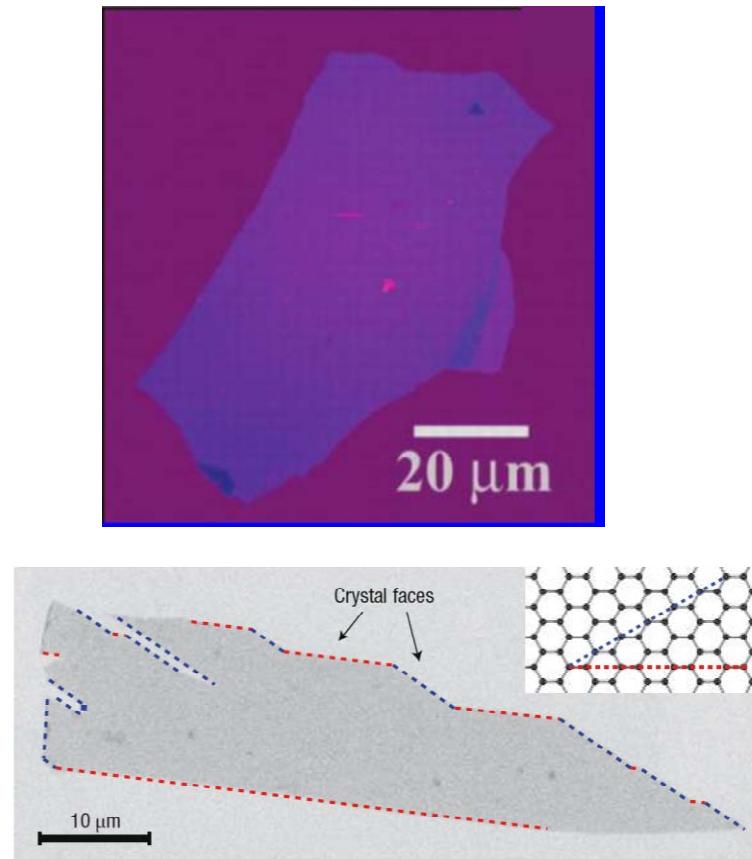


**How to find the atomic layer “graphene” from
repeatedly split graphite crystals by adhesive tape**

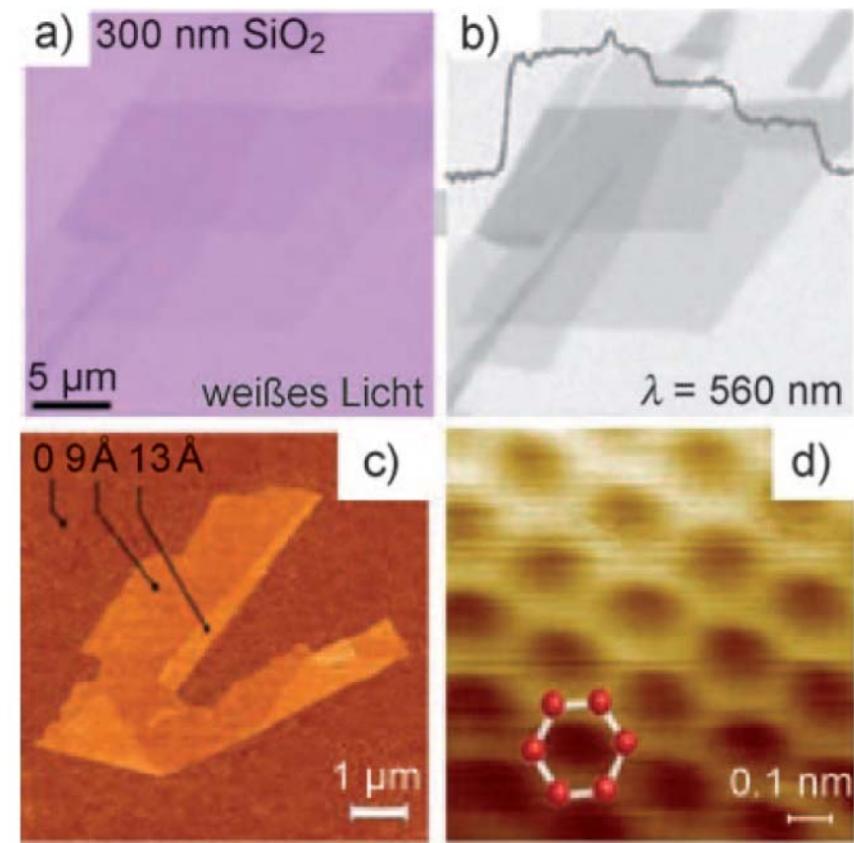


The properties of graphene

- The atomic structure, two-dimensional crystals
- The thinnest Materials

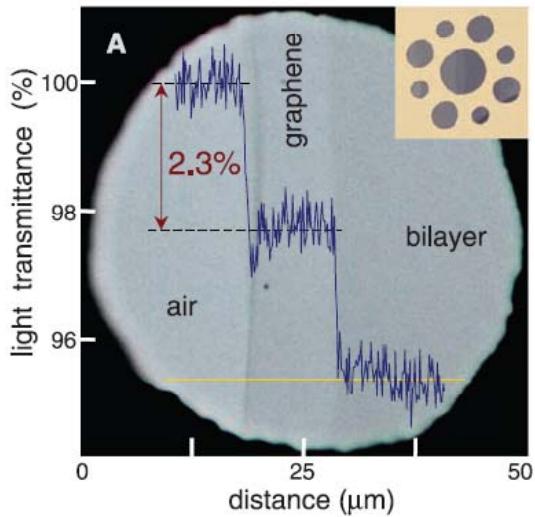
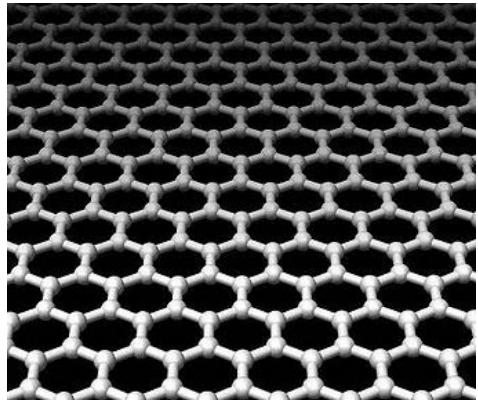


Kraner et al. Chem. Rev. 2010, 110, 132

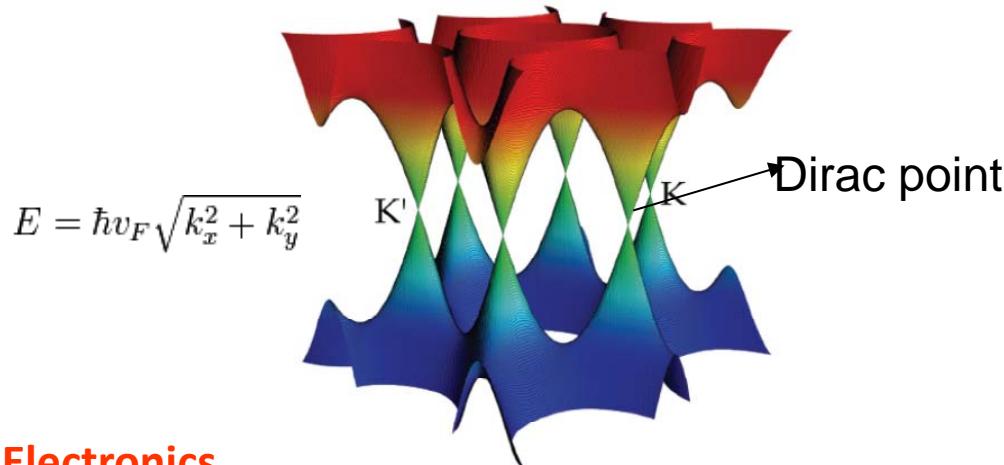


Rao et al, Angew. Chem., 2009, 48, 7752

Graphene



Nair et al., Science, (2008)



Electronics

- Zero effective mass near the Dirac point
- High carrier mobility $>15,000 \text{ cm}^2/\text{V}\cdot\text{s}$
- Low Resistivity about $10^{-6} \Omega\cdot\text{cm}$, (< silver)
- etc...

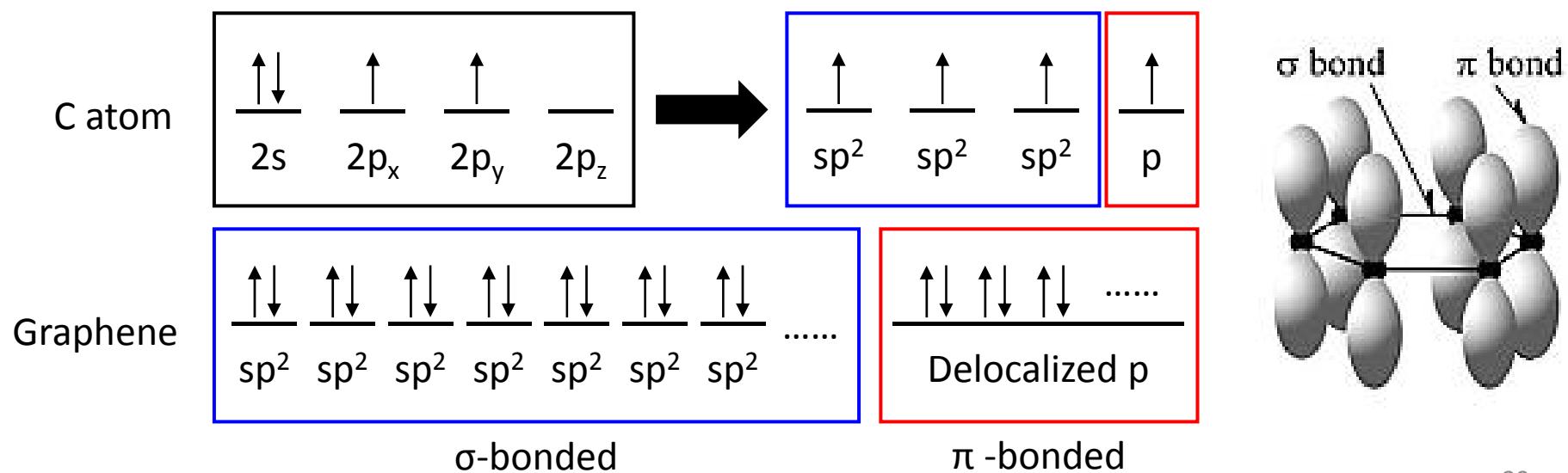
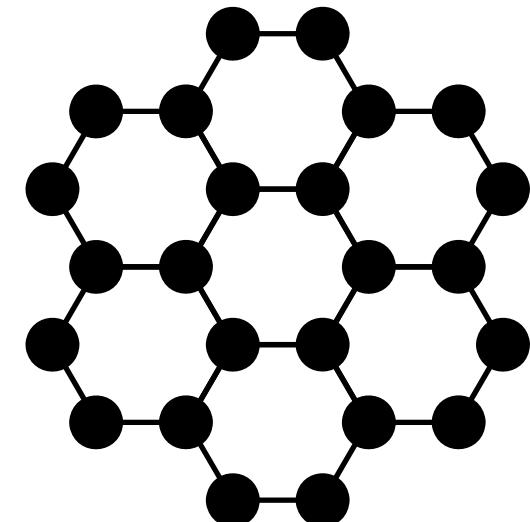
Optics

1. One atomic layer absorption $\frac{\pi e^2}{\hbar c} = \pi \alpha = 2.3\%$
2. High transparency

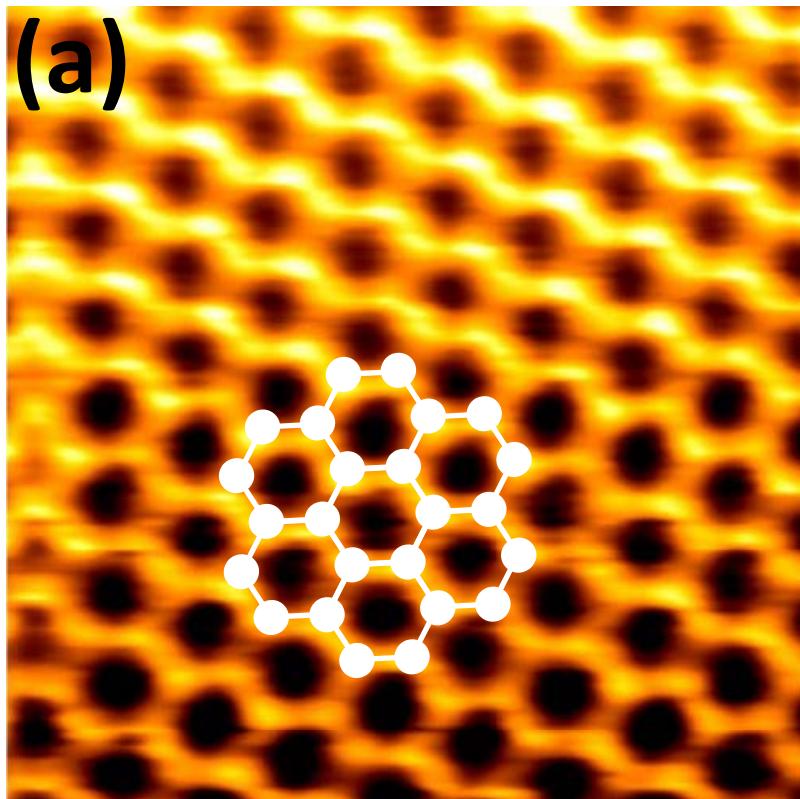
2010 Nobel prize award in Physics

Electronic Structure of Graphene

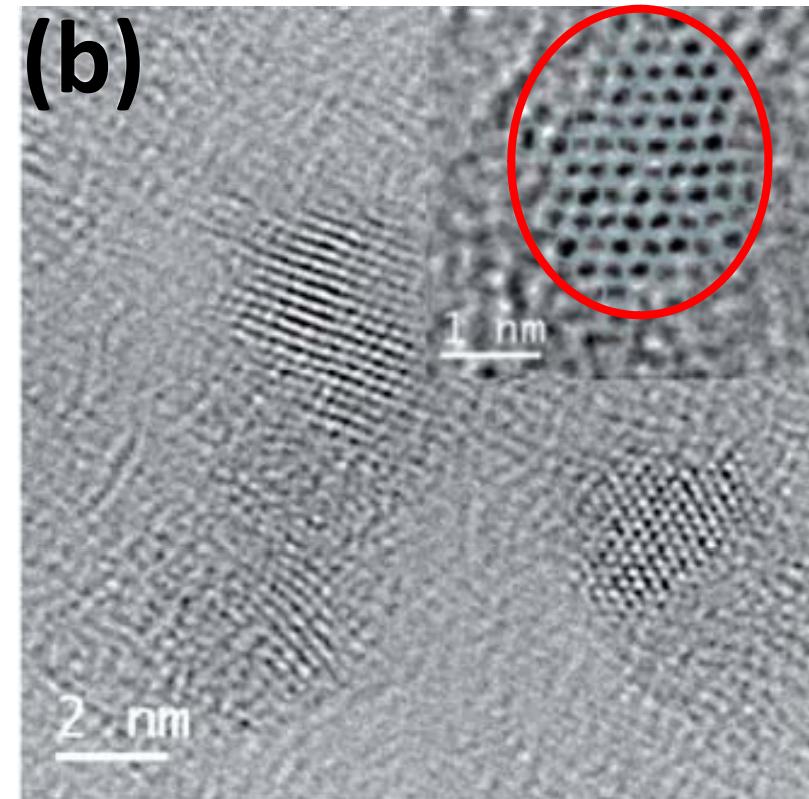
- All C atoms are sp^2 -bonded to adjoining C atoms
 - sp^2 electrons form σ bonds
 - Form the honeycomb net of C atoms
 - Delocalized p electrons form π bonds



Electronic Structure of Graphene

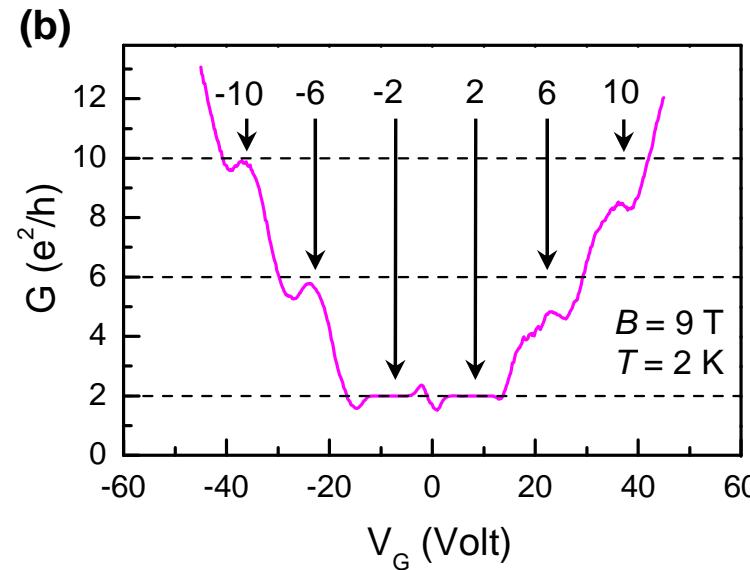
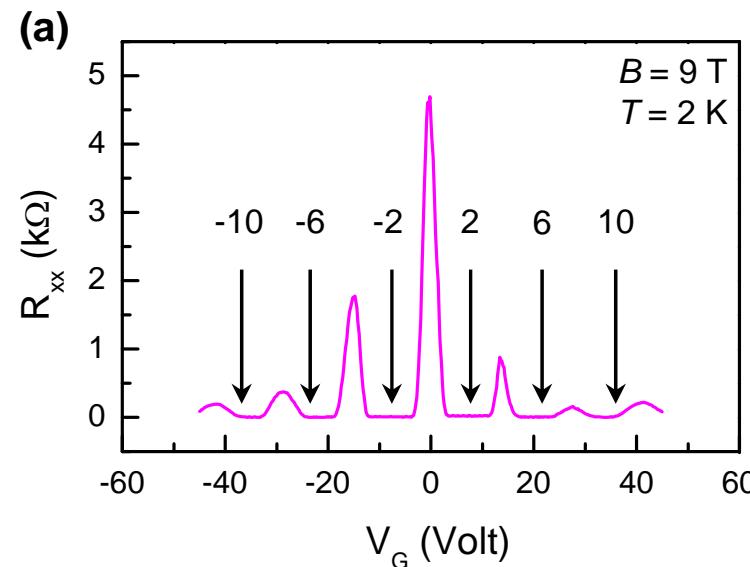
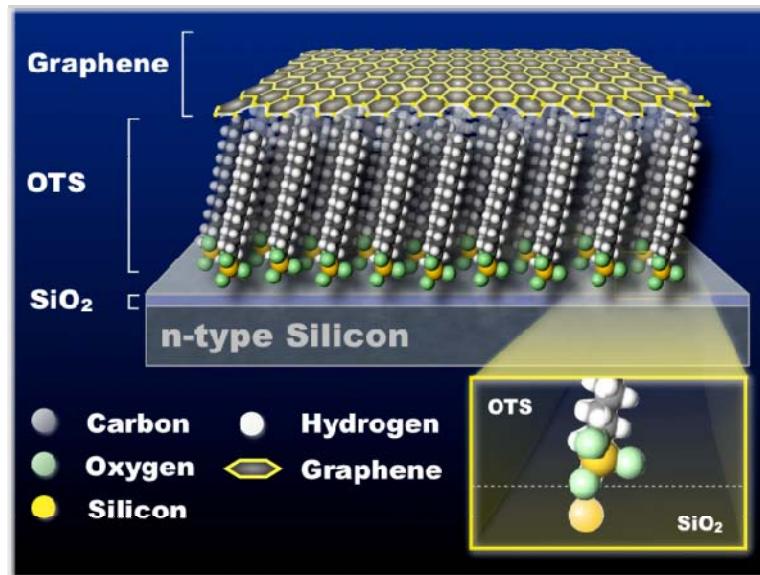
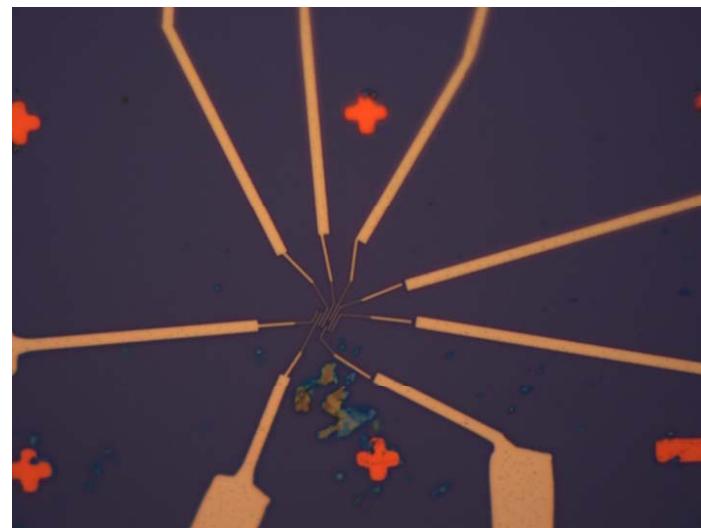


(a) STM image of graphene



(b) TEM image of graphene
clusters within graphene oxide

Graphene transistor device with high mobility

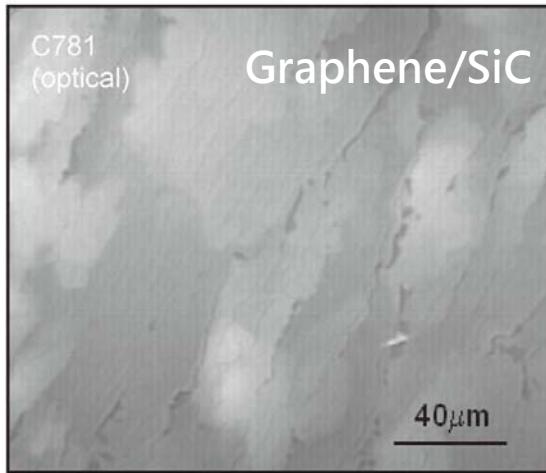


Nano Letters, 12, 964-969 ,(2012)

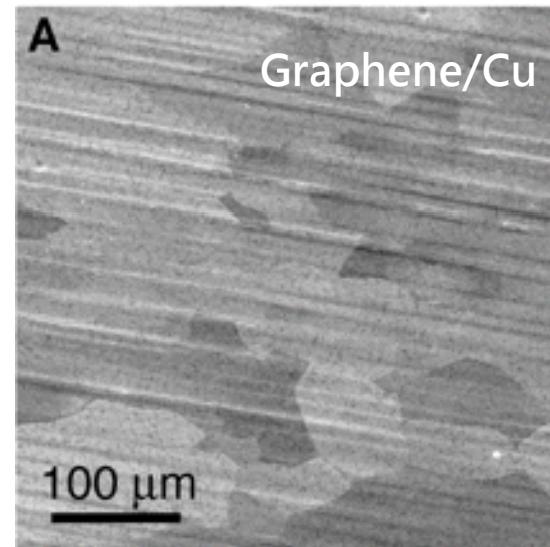
1-2. Synthesis of Graphene

Synthesis of Graphene

- Mechanical exfoliation
- Epitaxial growth on silicon carbide
- Epitaxial growth on metal substrates
- Reduction of graphene oxide
=> solution processible, mass producible,
simple and cheap



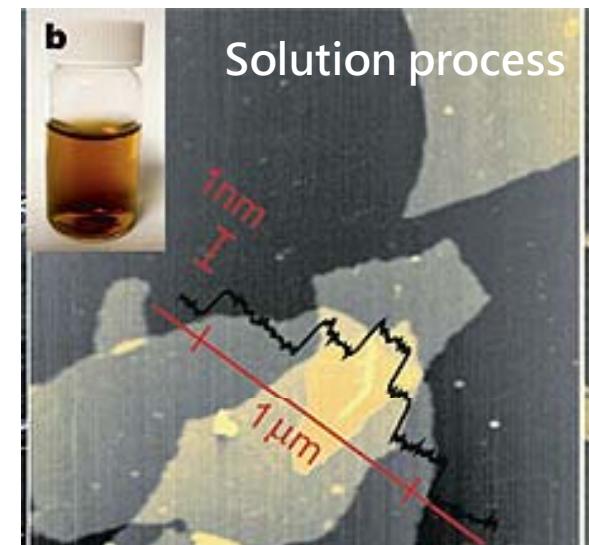
de Heer et al, *Science* (2006)



Ruoff et al., *Science*, vol. 324,
pp1312-1314, 2009



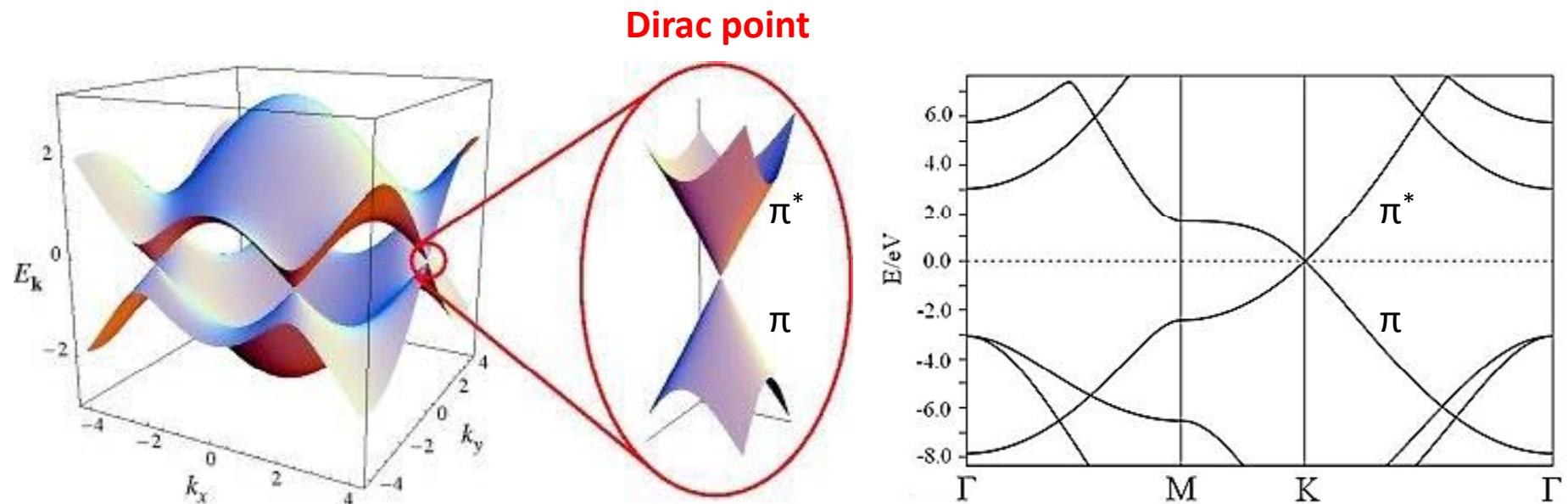
Novoselov et al., *Science* (2004)



Sasha Stankovich, et al., *Nature* 442,
282-286, 2006

1-3. Properties of Graphene

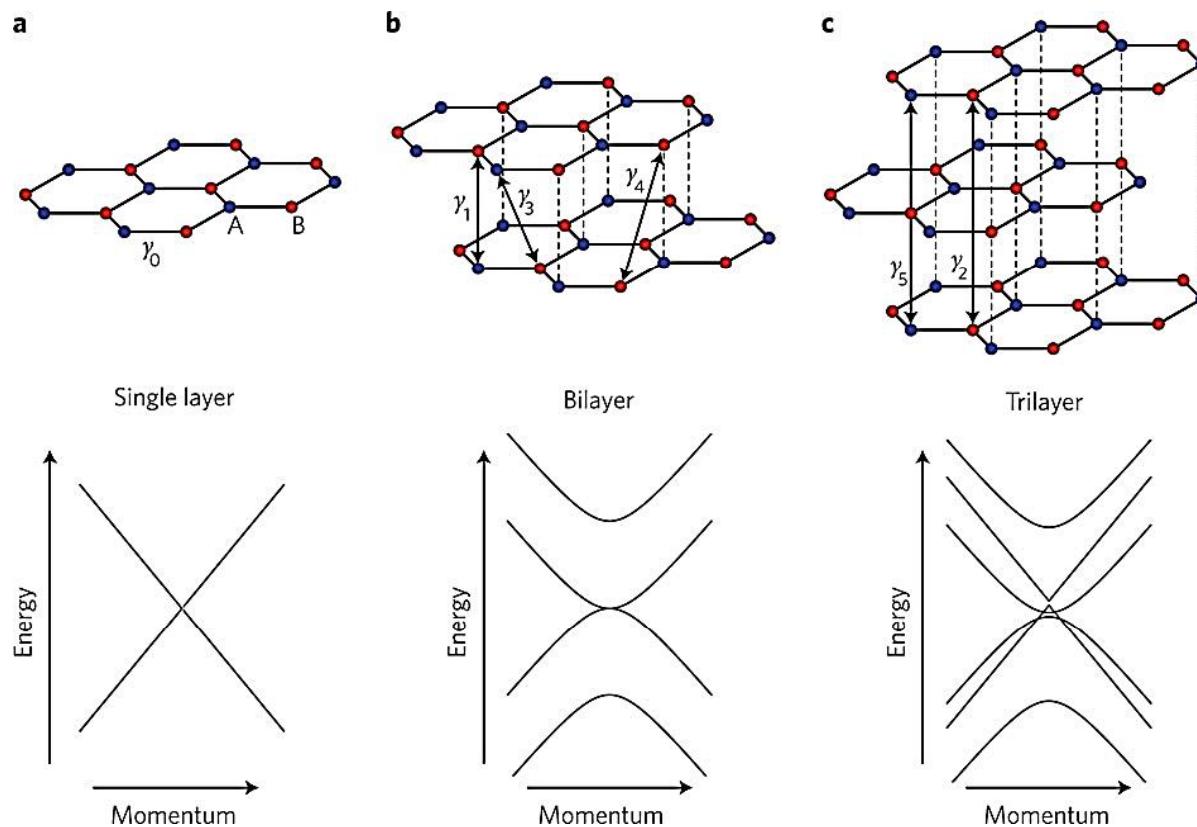
Band Structure of Graphene



- The valence band and the conduction band meet at Dirac point
 - Metallic behavior
 - “Semi-metal” or “zero-bandgap semiconductor”
- Linear E-k dispersion near Dirac point
 - “Massless” electrons and holes

Band Structure of Graphene

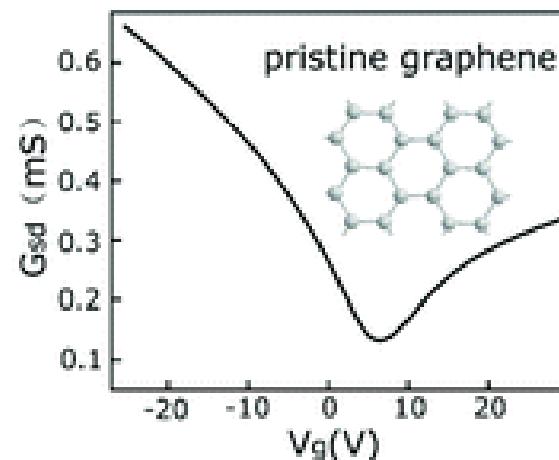
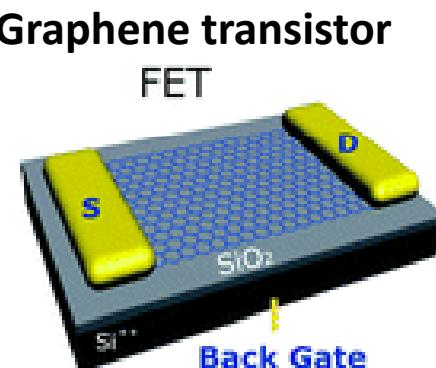
- How about multi-layered graphene?



[8] Marcus Freitag, *Nature Physics* **7**, 596 (2011)

Electronic Properties

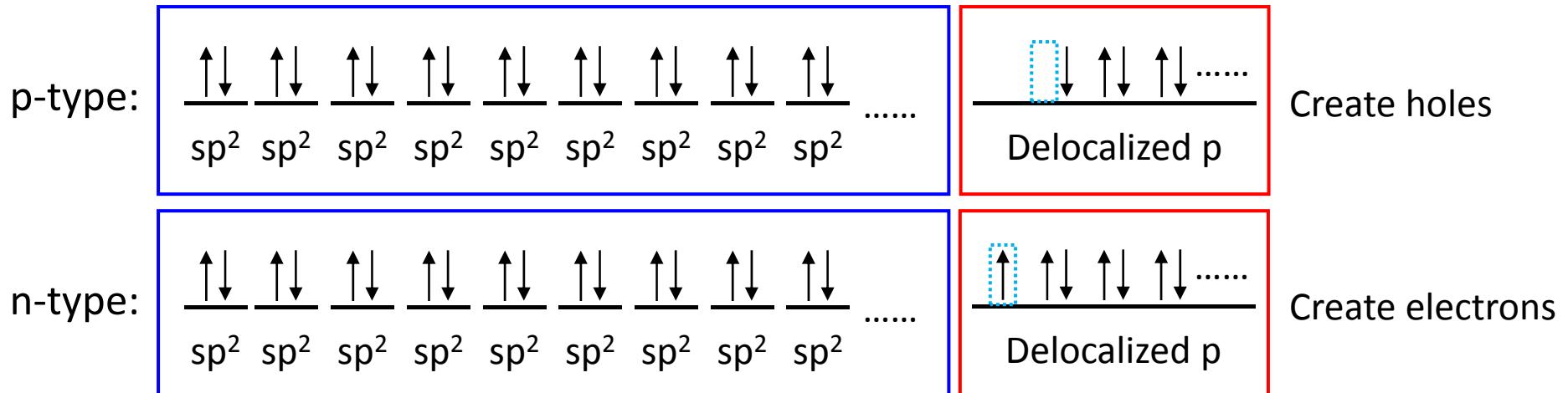
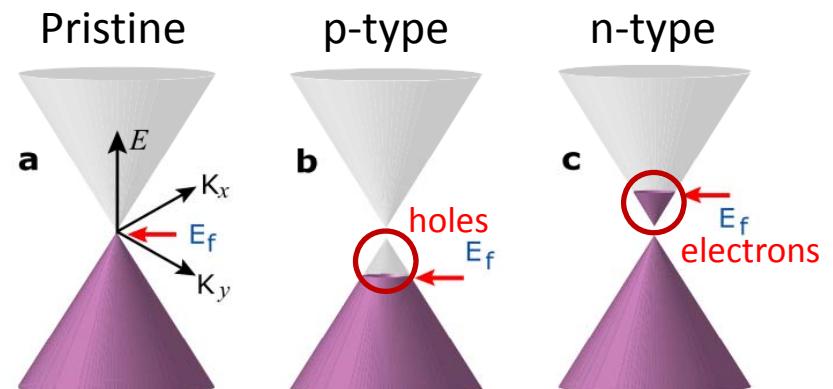
- Semi-metal
 - Zero bandgap
- High electron mobility at room temperature
 - Excess of $15,000 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$
- Can be doped
 - In order to increase the carrier concentration



Nano Lett., 2010, 10 (12), pp 4975–4980

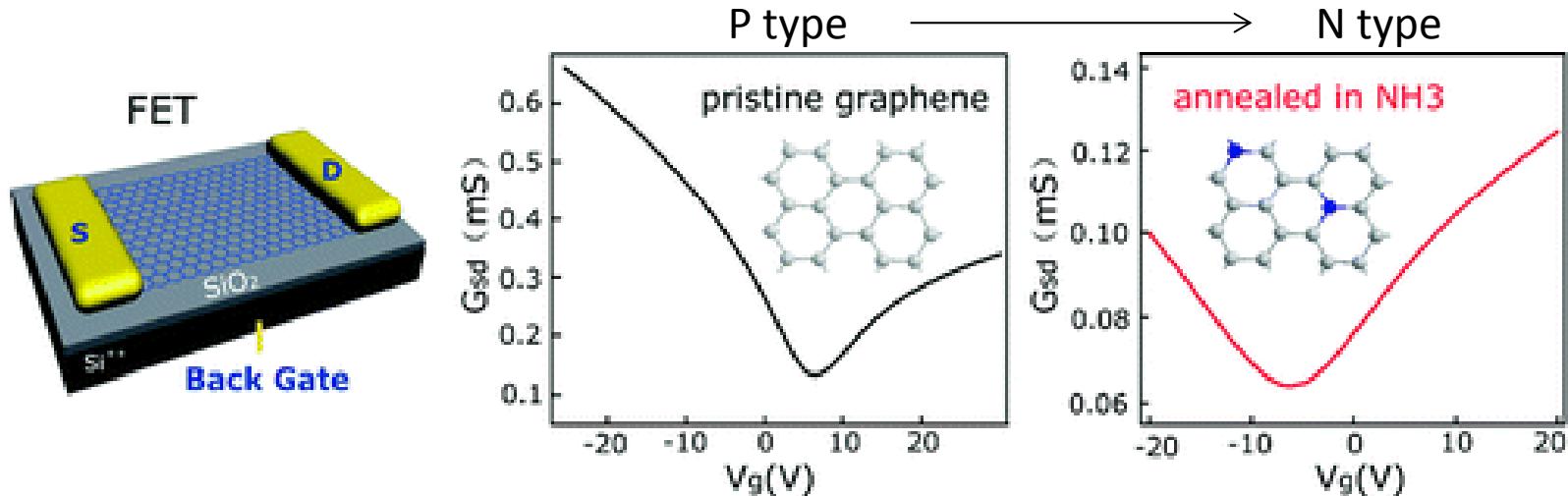
Doped Graphene by heteroatoms

- Doping by heteroatoms
 - p-type doping
 - Group IIIA, such as B
 - n-type doping
 - Group VA, such as N and P

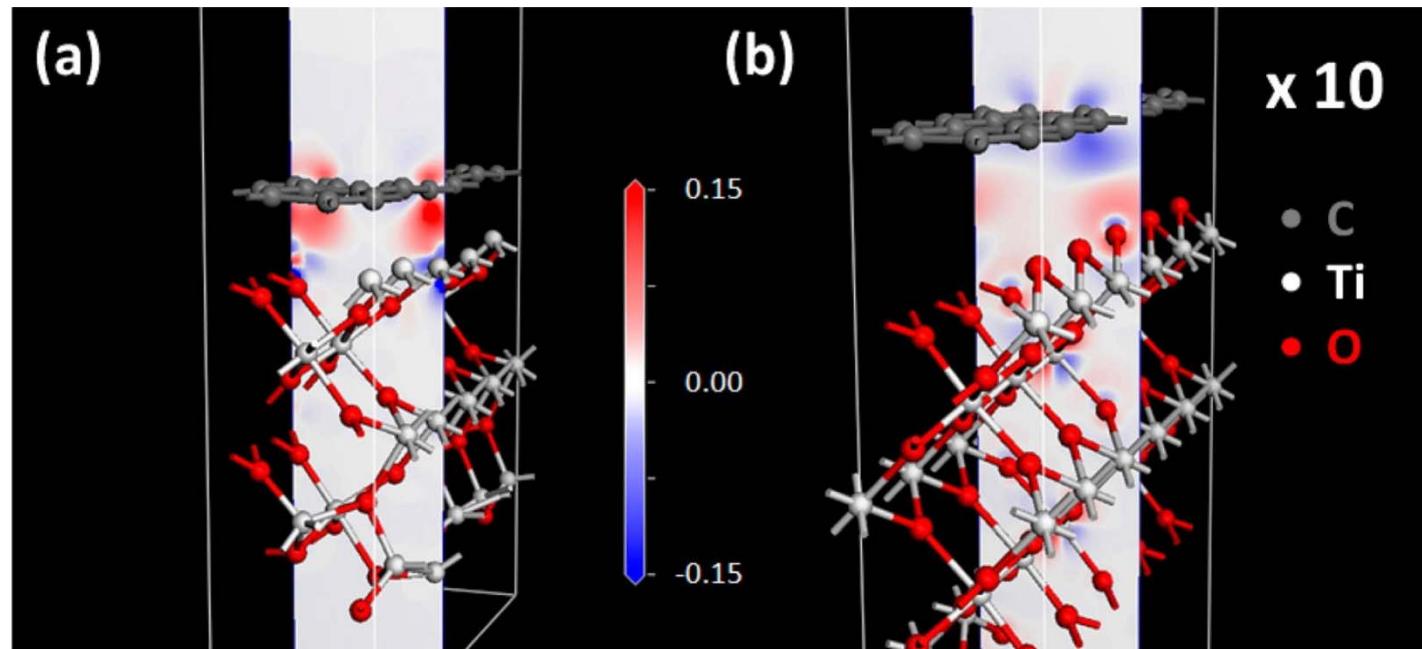


Doped Graphene by chemical modification

- Doping by chemical modification
 - Charge transfer
 - Molecules adsorb on graphene, acting as donors or acceptors
 - Epitaxial graphene can be doped by the substrate



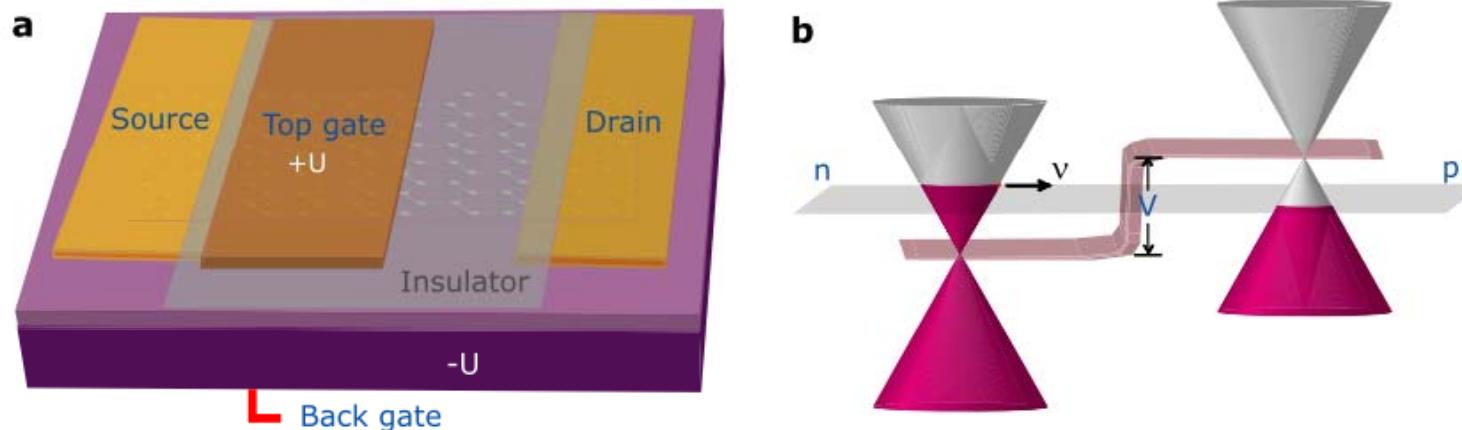
Substrate Doping of Graphene



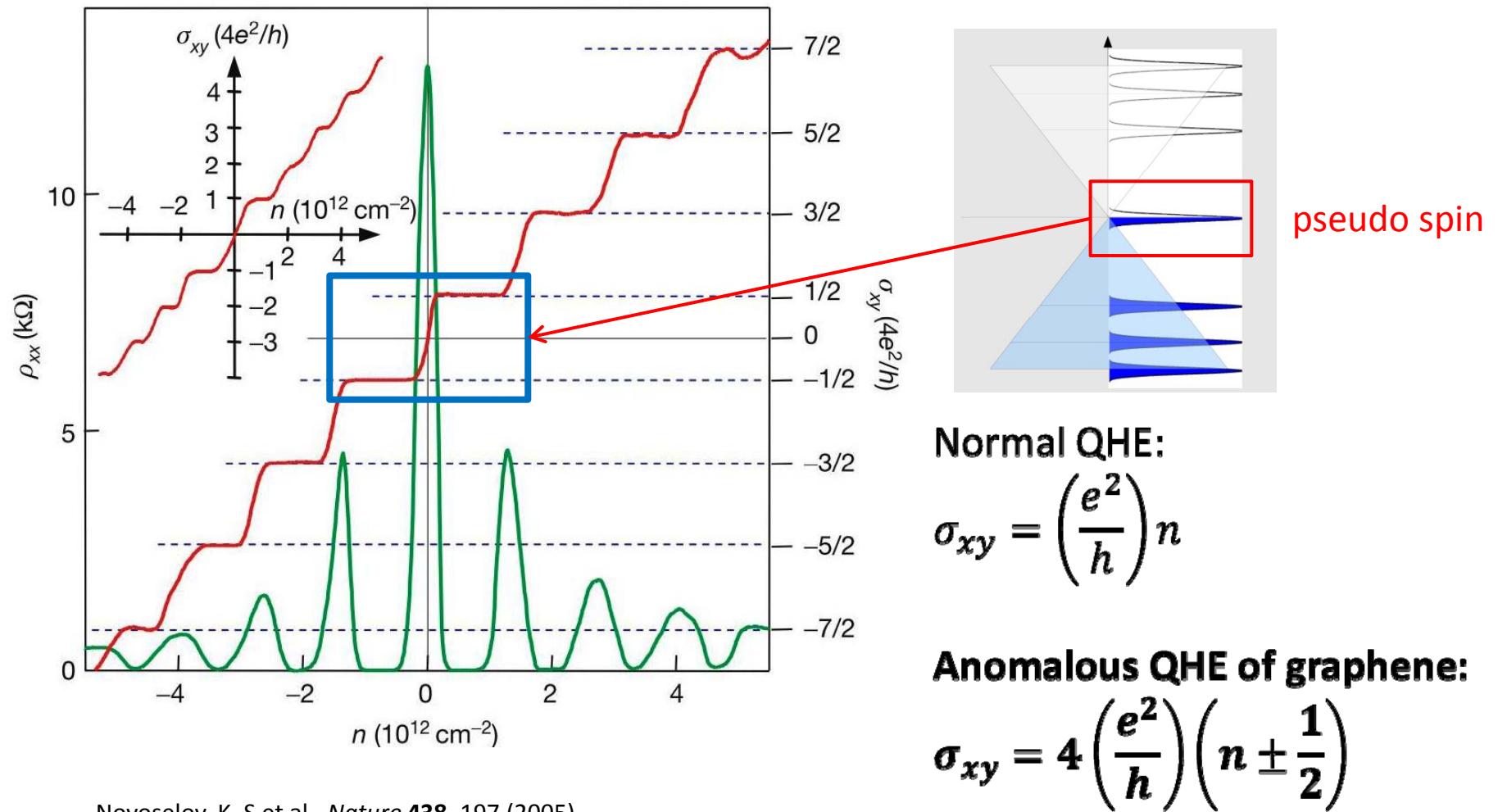
- (a) Graphene is n-type doped by Ti-rich TiO_2 surface
- (b) Graphene is p-type doped by O-rich TiO_2 surface

Doped Graphene by electric field

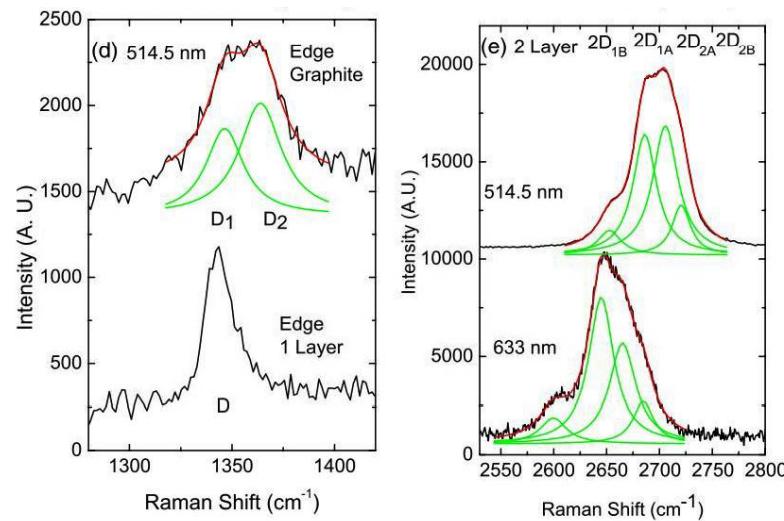
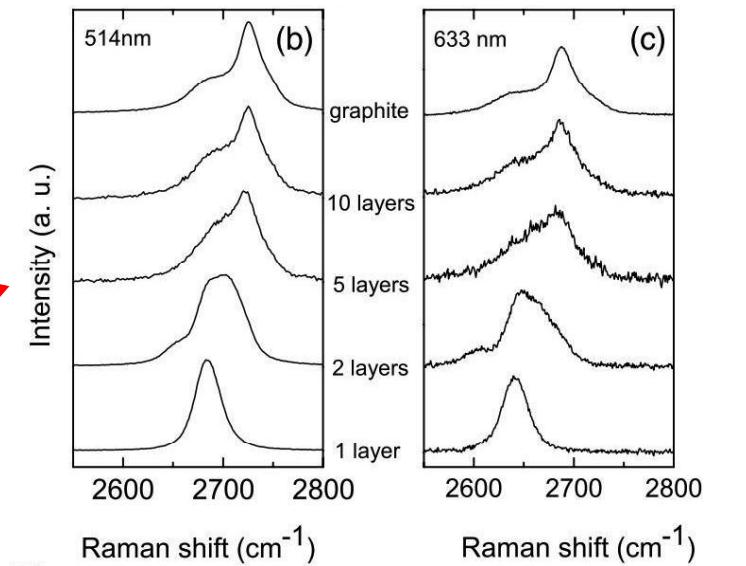
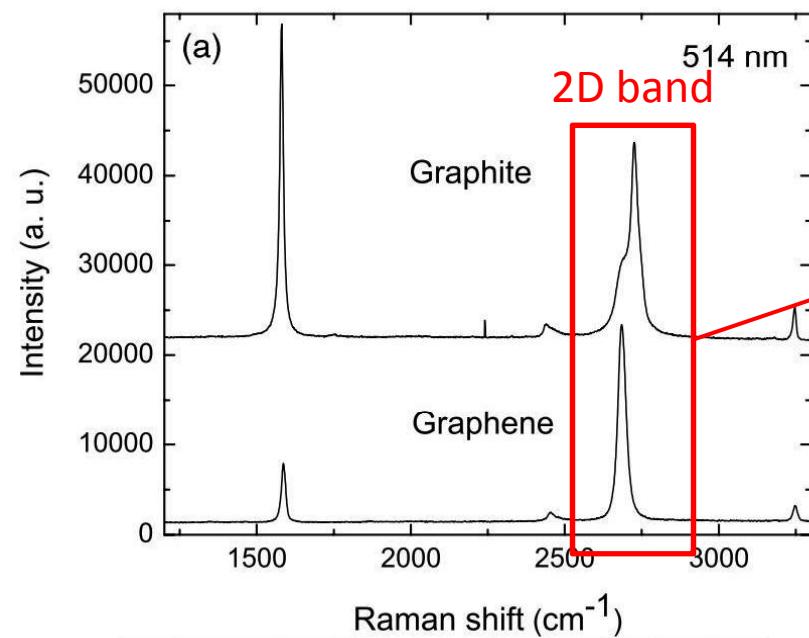
- Doping by electric field
 - Use electric field to shift the Fermi level of graphene



Anomalous Quantum Hall Effect



Raman Spectra of graphene

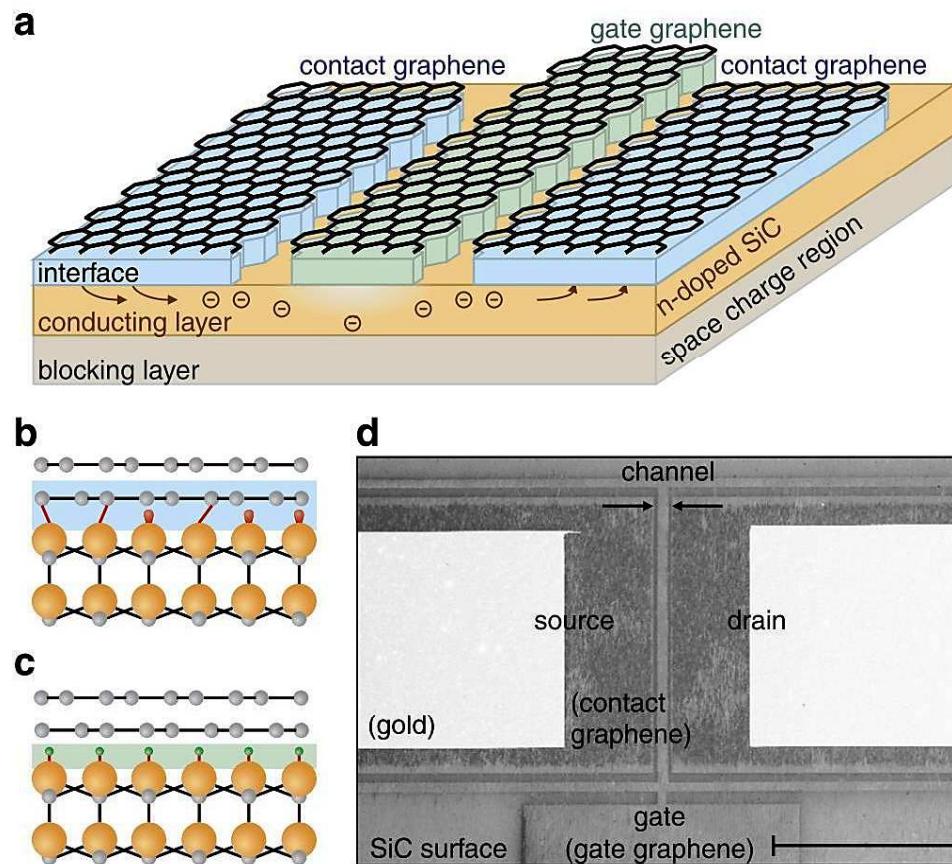


- Identify the layers of graphene

A. C. Ferrari et al., PRL 97, 187401 (2006)

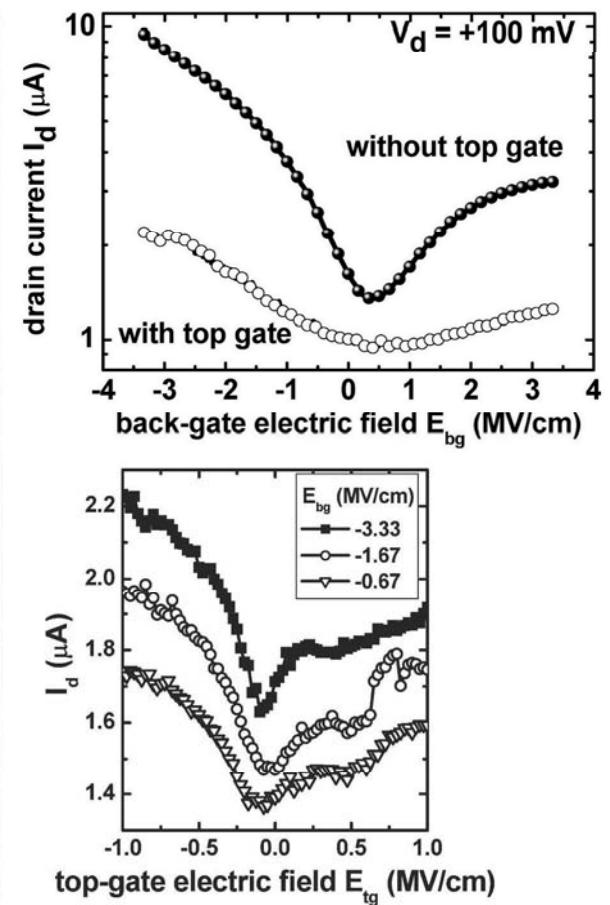
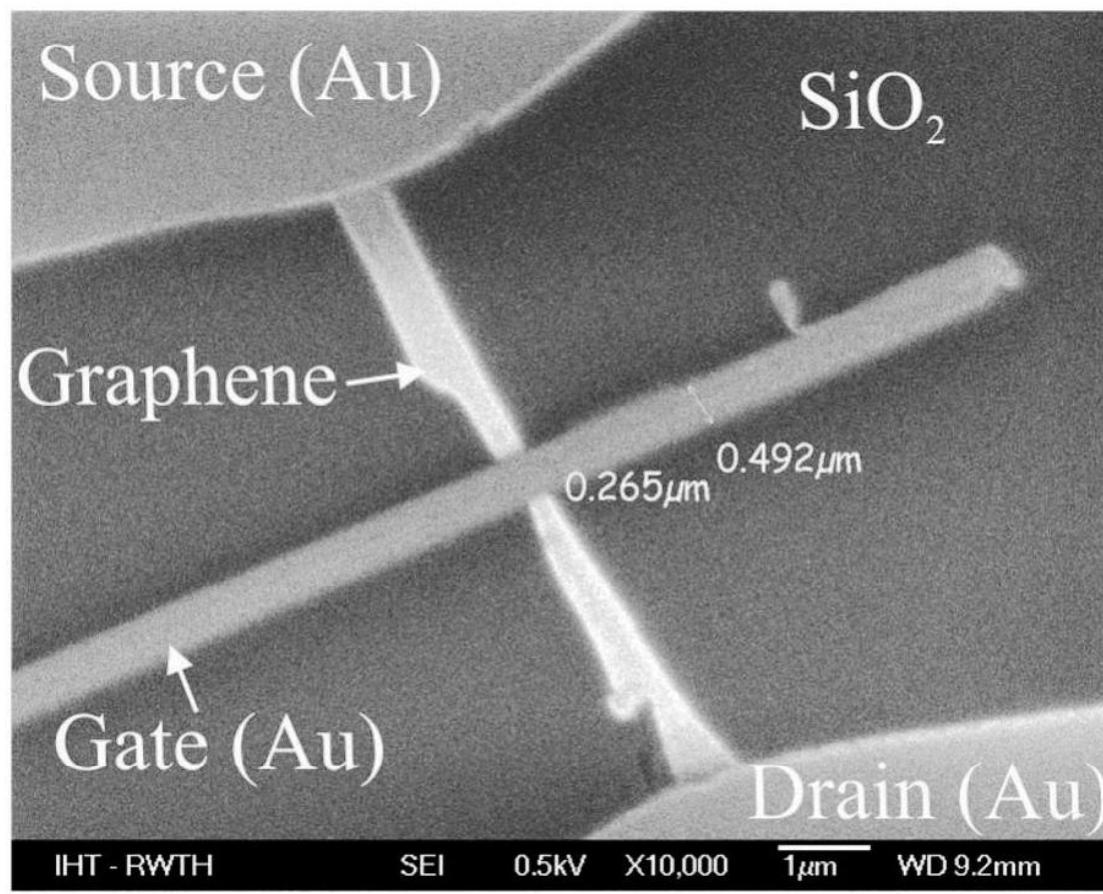
1-4. Applications of Graphene

Graphene Transistor



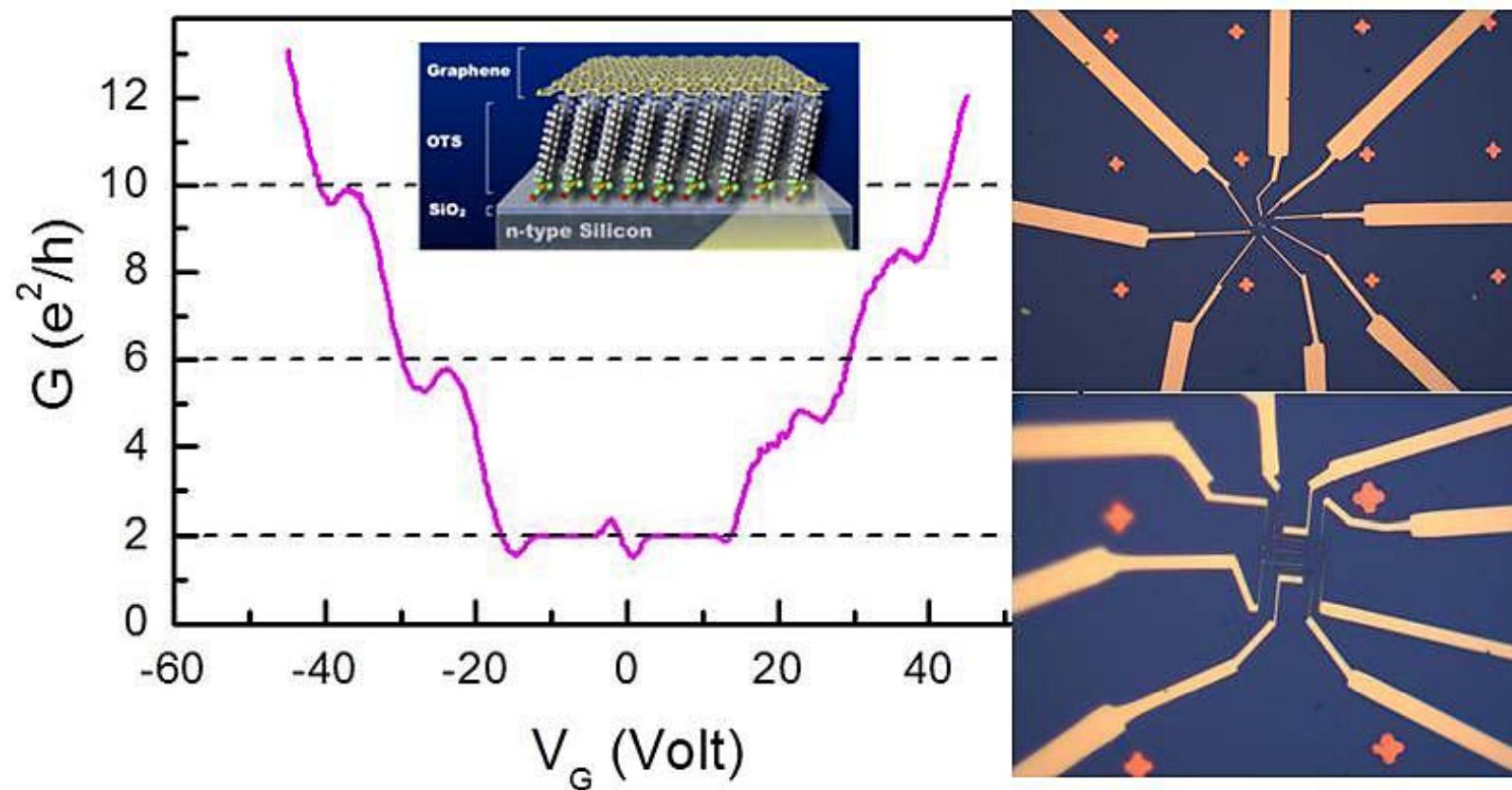
[20] S. Hertel et al., *Nat. Commun.* **3**, 957 (2012)

Graphene Transistor



Max C. Lemme et al., *EDL* **28** (4), 282 (2007)

Graphene Transistor



[22] *Nano Lett.* **12** (2), 964 (2012)

Next generation production ?



© NOKIA



Household appliances

E-paper

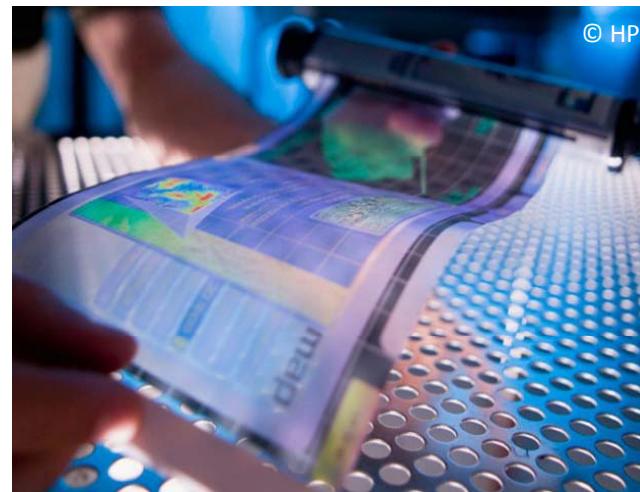


- High conductivity
- High transparency
- Good mechanical property
- Flexible
- Good thermal conductivity
- etc.



© SAMSUNG

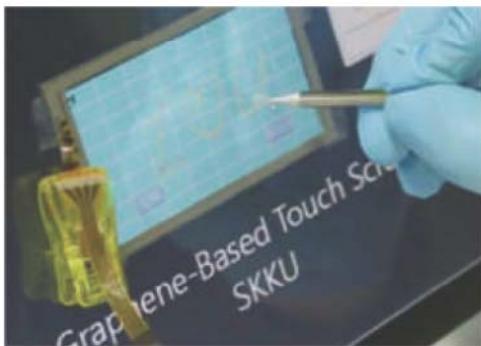
Communication
production



© HP

- Transparent conducting electrode

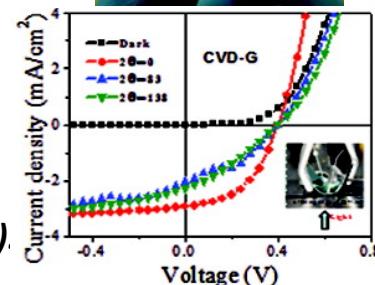
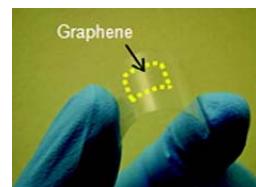
Touch Panel



Bae, S. et al.

Nature Nanotech. **4**, 574–578 (2010)

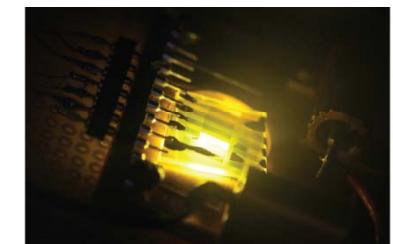
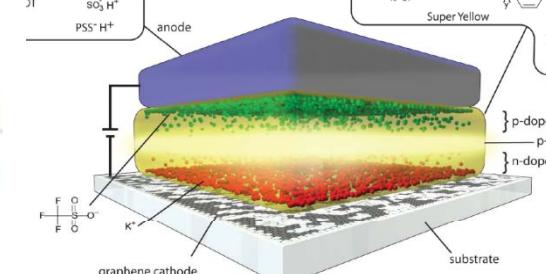
Solar Cell



De Arco, L. G. et al

ACS Nano **4**, 2865–2873 (2010)

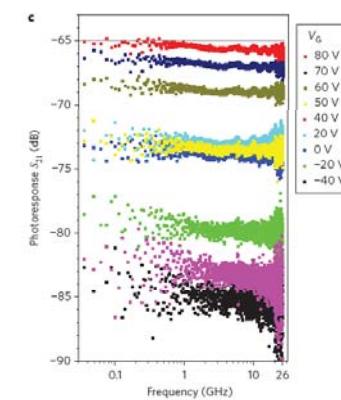
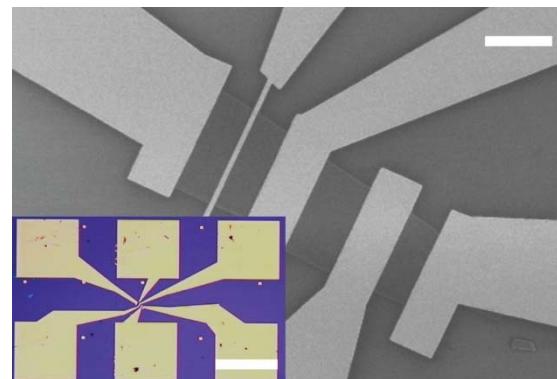
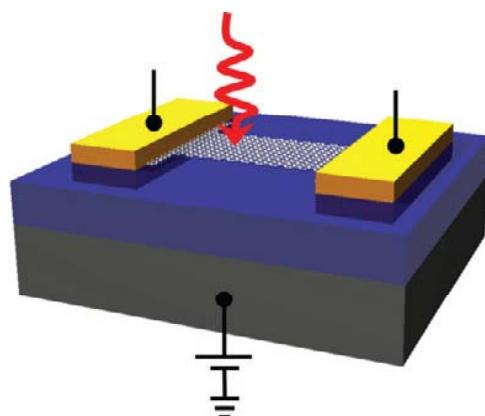
Light Emitting Diode



Matyba, P. et al.

ACS Nano **4**, 637–642 (2010).

- Ultra fast Photodetector



59

Xia, F. et al. *Nature Nanotech.* **4**, 839–843 (2009).

Large area CVD grown graphene for optoelectronic applications

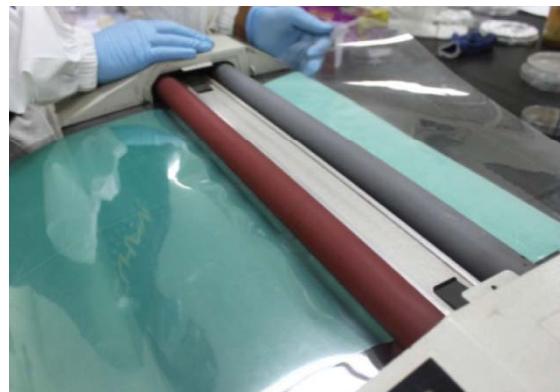
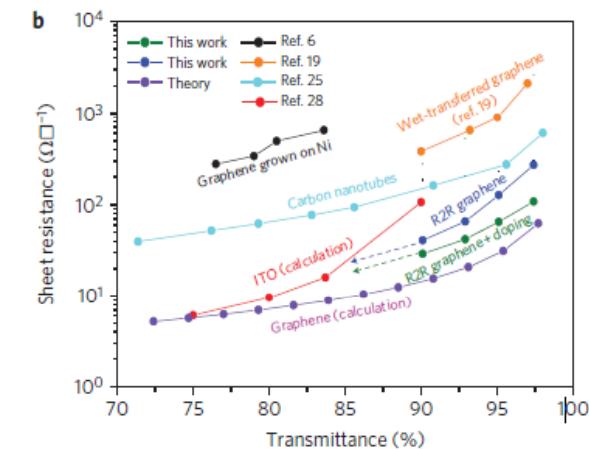
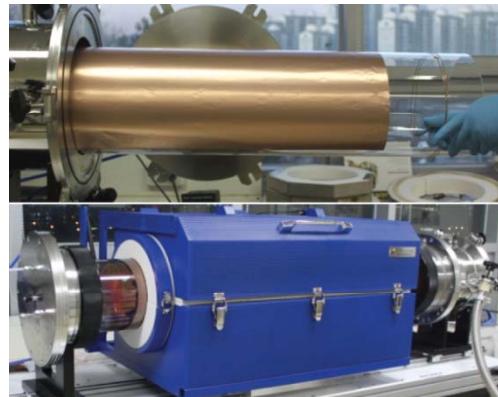
nature
nanotechnology

LETTERS

PUBLISHED ONLINE: 20 JUNE 2010 | DOI: 10.1038/NNANO.2010.132

Roll-to-roll production of 30-inch graphene films for transparent electrodes

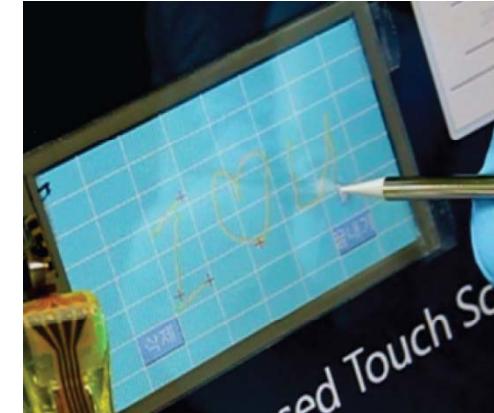
Jong-Hyun Ahn and Byung Hee Hong *et al.**



Roller printing



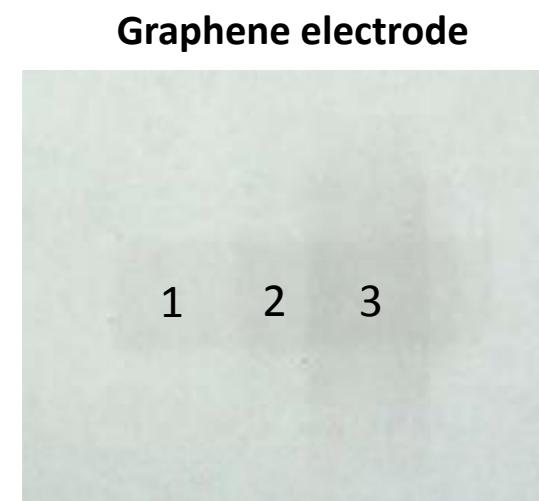
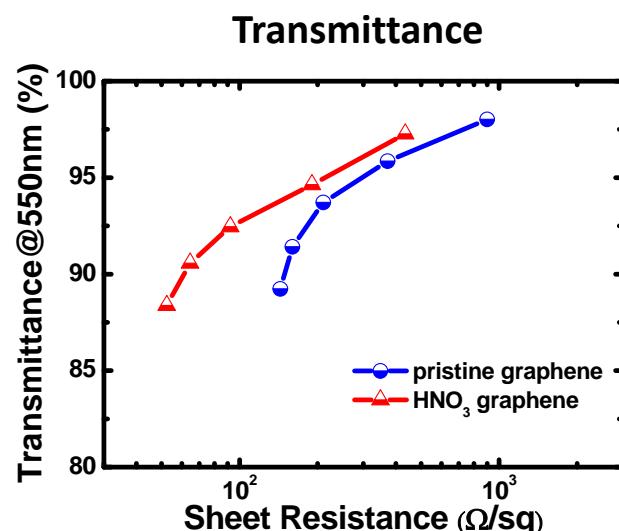
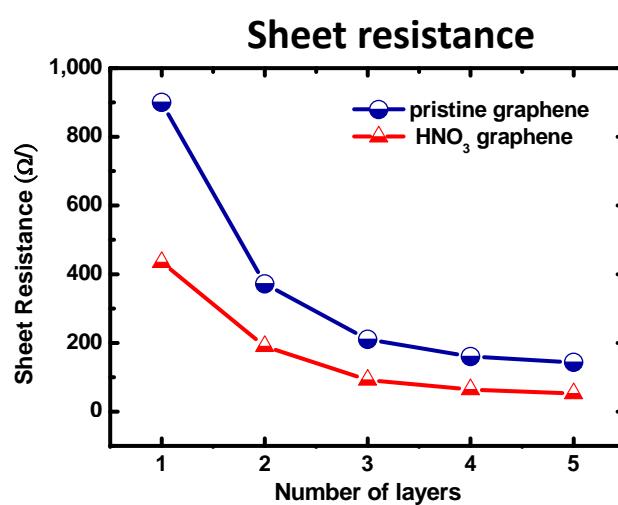
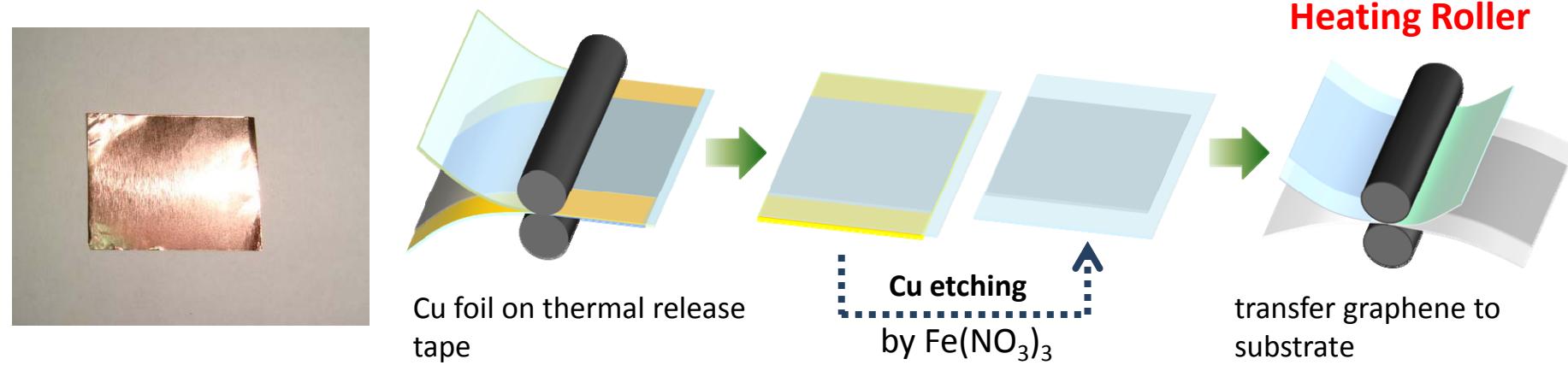
Scale up



Touch panel

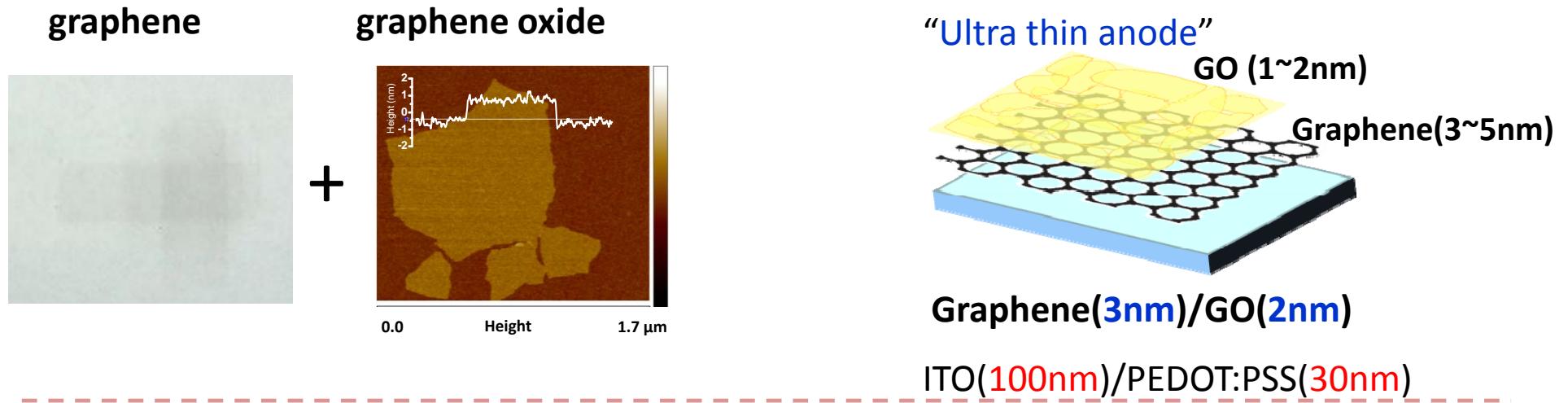
Fabrication of CVD Graphene Electrode on PET

Graphene on Cu foil by CVD



($\sim 55\Omega/\text{sq}$) at a transparency. ($\sim 90\%$)

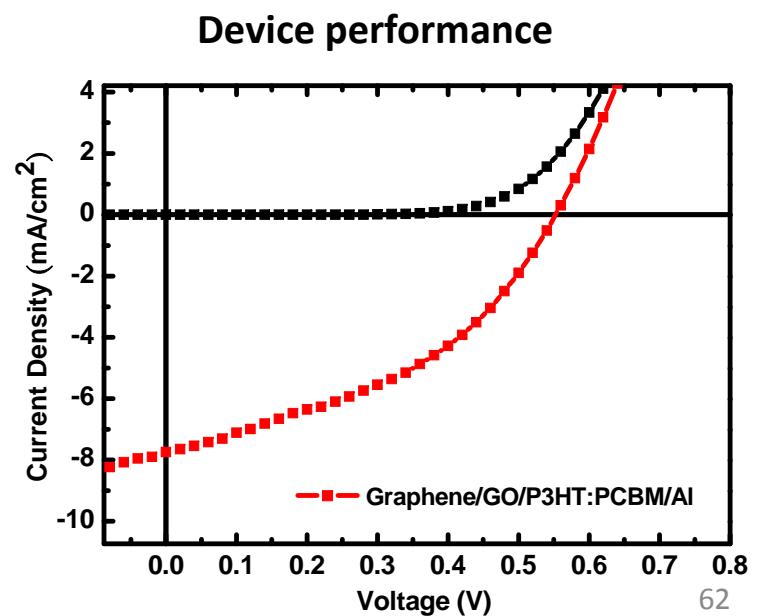
Polymer solar cells based on “Ultra thin” Graphene/GO anode by Roll-to-roll technology on PET



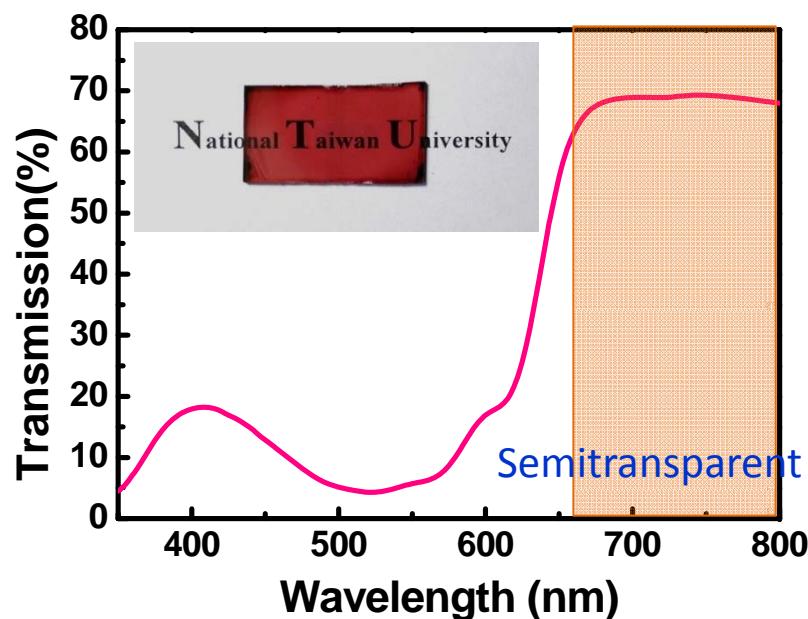
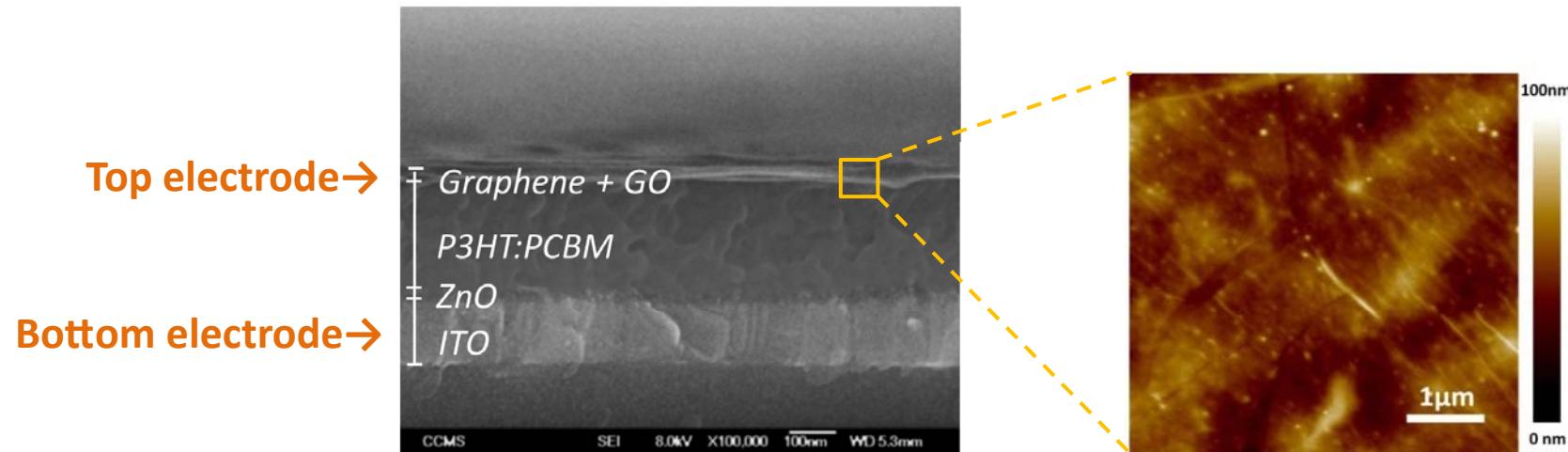
Polymer solar cell based on “graphene” platform !



Graphene/GO/P3HT:PCBM/AI

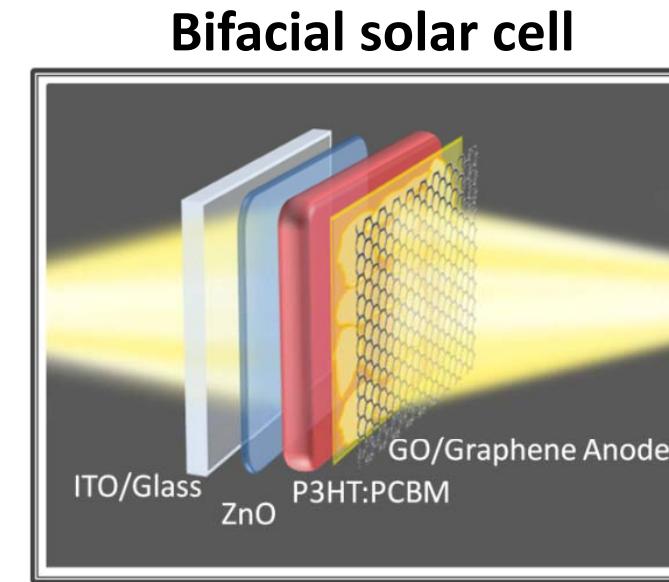
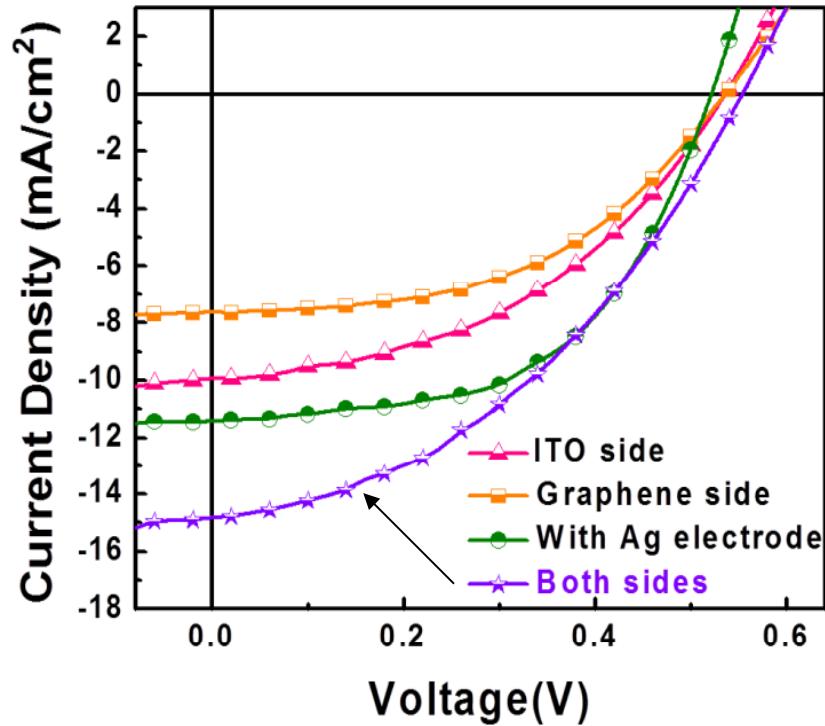


Top laminated graphene electrode in a semitransparent polymer solar cell



ACS Nano , Vol.5, 6564, (2011)

Device performance of a bifacial polymer solar cell using graphene top electrode



ACS Nano , 5, 6564, (2011)

| | V_{oc} (V) | J_{sc} (mA/cm^2) | FF | PCE(%) |
|----------------------|--------------|--------------------------------------|------|--------|
| Graphene side | 0.54 | 7.7 | 0.49 | 2.04 |
| ITO side | 0.54 | 10.1 | 0.44 | 2.50 |
| Ag electrode | 0.52 | 11.5 | 0.55 | 3.30 |

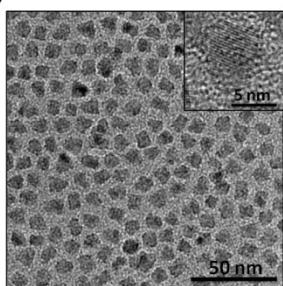
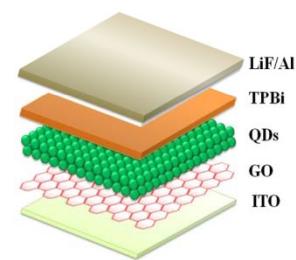
~75% of a standard opaque device

Graphene-based materials for Polymer and QD LED



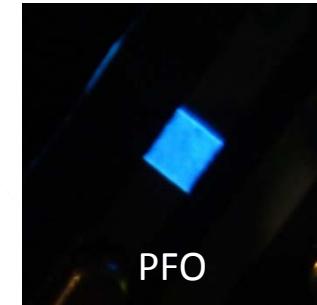
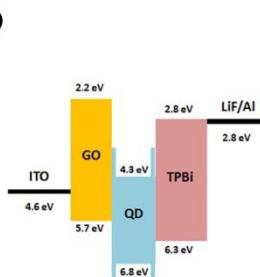
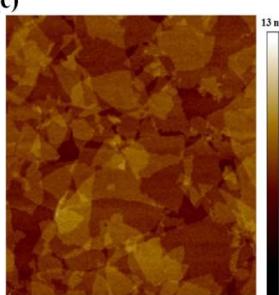
(a)

(b)

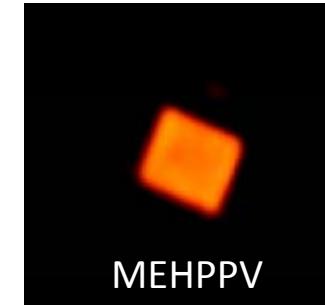


(c)

(d)



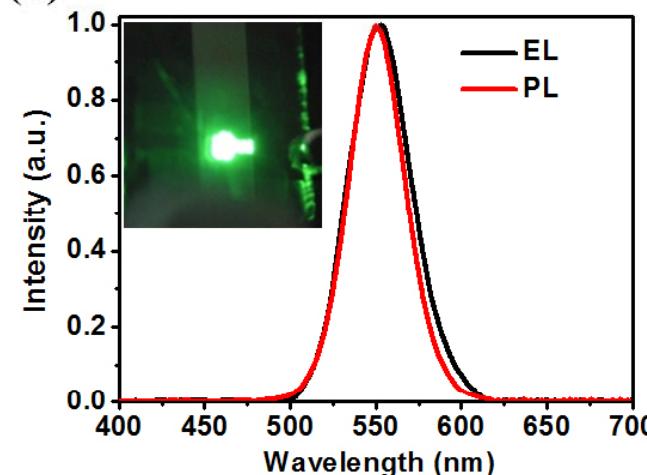
PFO



MEHPPV

Polymer LED

QD LED



Other optoelectronics application of Graphene

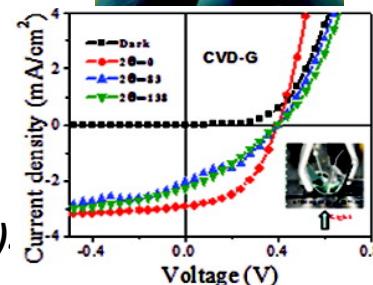
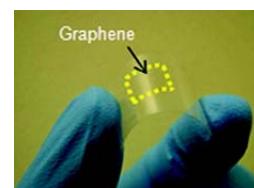
- Transparent conducting electrode

TOUCH PANEL



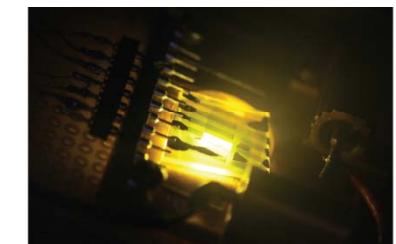
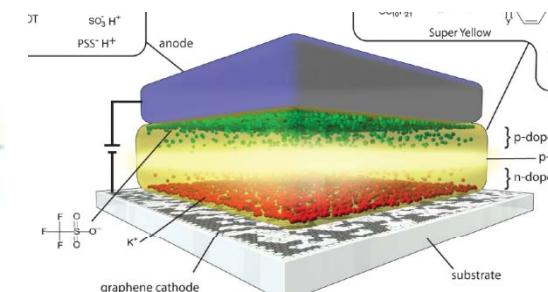
Bae, S. et al.
Nature Nanotech. 4, 574–578 (2010)

SOLAR CELL

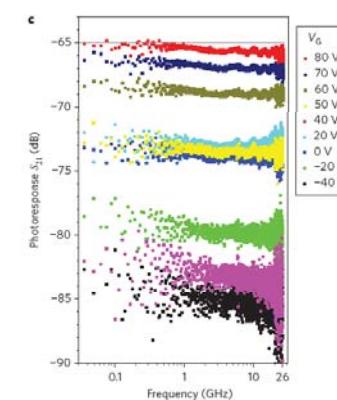
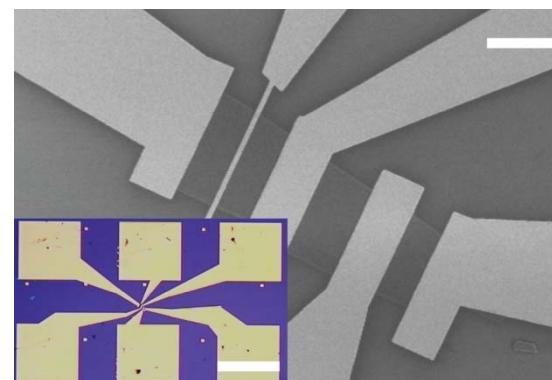
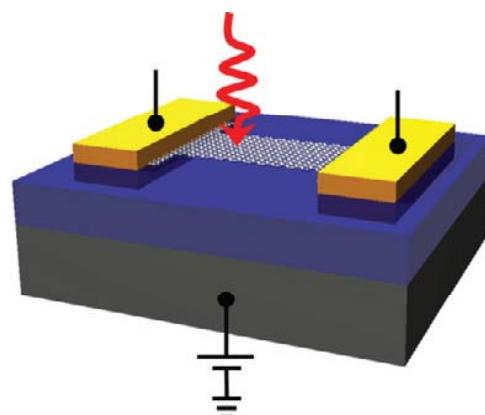


De Arco, L. G. et al
ACS Nano 4, 2865–2873 (2010)

Light emitting diode



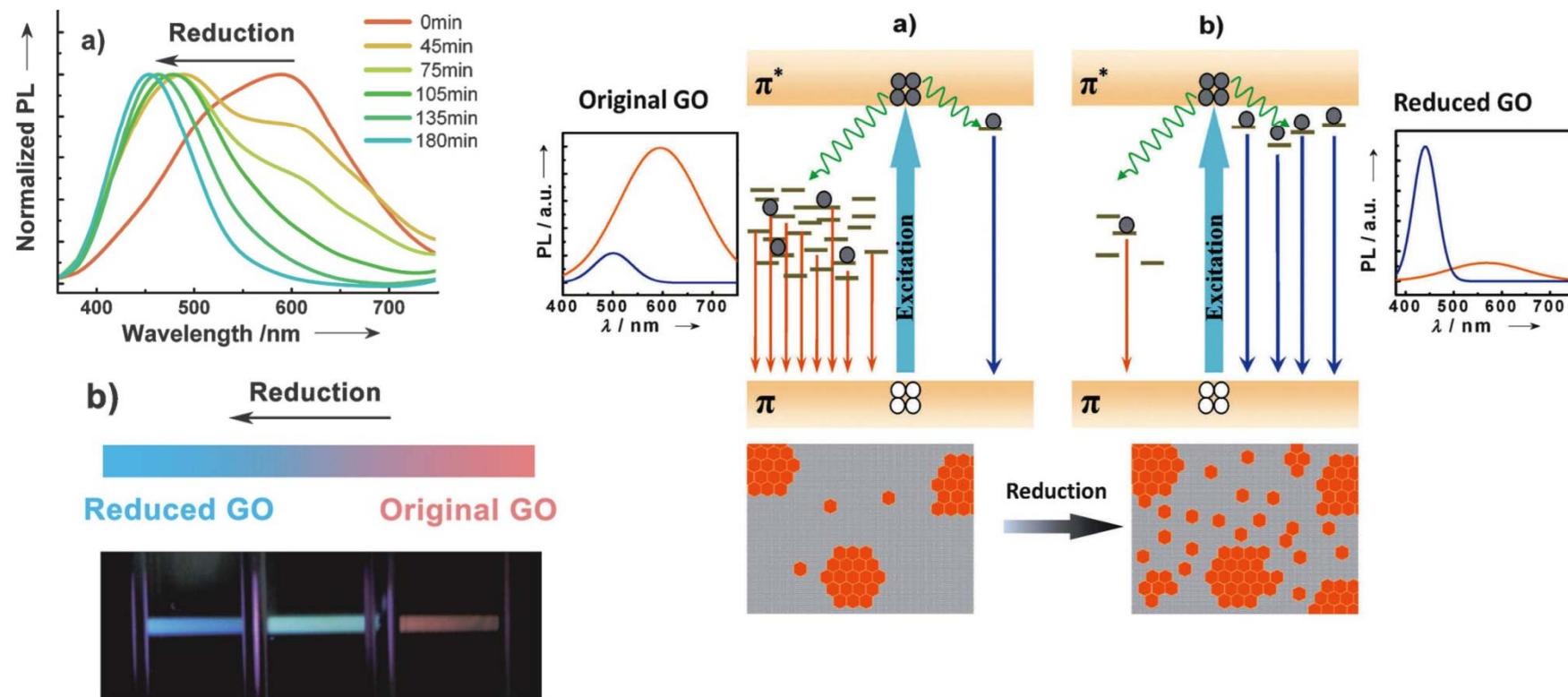
Matyba, P. et al.
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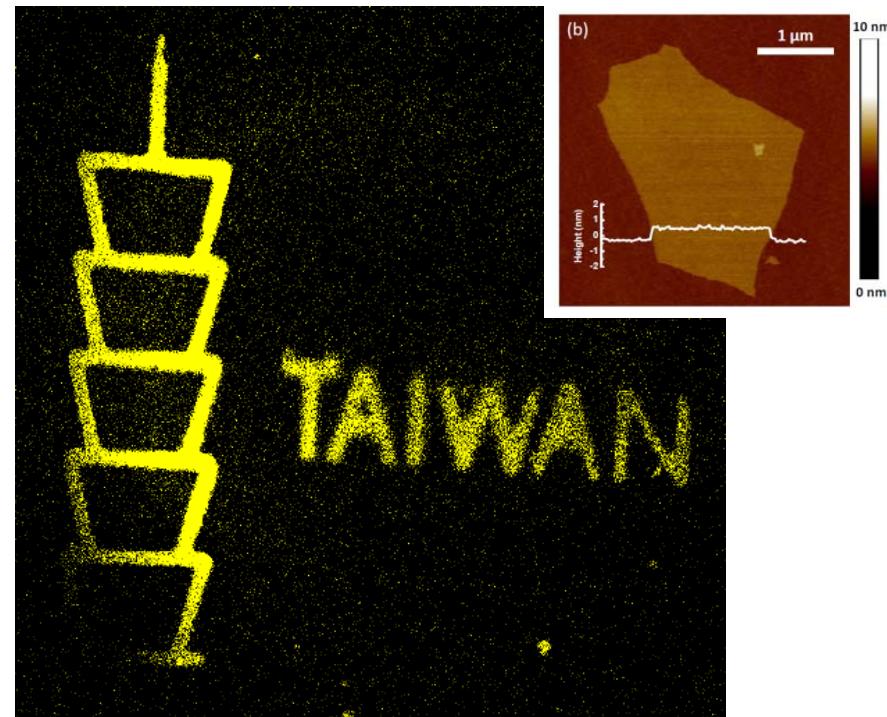
Xia, F. et al. *Nature Nanotech.* 4, 839–843 (2009).

Tunable PL of Graphene /GO



CT Chien, CW Chen et al., *Angew. Chem. Int. Ed.* **51** (27), 6662 (2012)

石墨烯氧化物之發光



Advanced Materials, 22, 505, (2010)
Angewandte Chemie, (2012)

End of Part II